



# Carbon: Seeing the Forest (and Products) for the Trees

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Group 1 IA												18 VIII															
1	<b>H</b> Hydrogen (1.00784, 1.00811)											2	<b>He</b> Helium (4.002602(2))														
2	<b>Li</b> Lithium (6.938, 6.997)	<b>Be</b> Beryllium (9.0121831(5))											<b>B</b> Boron (10.806, 10.821)	<b>C</b> Carbon (12.0096, 12.0116)	<b>N</b> Nitrogen (14.00643, 14.00728)	<b>O</b> Oxygen (15.99903, 15.99977)	<b>F</b> Fluorine (18.998403163(6))	<b>Ne</b> Neon (20.1797(6))									
3	<b>Na</b> Sodium (22.98976928(2))	<b>Mg</b> Magnesium (24.304, 24.307)											<b>Al</b> Aluminum (26.9815384(3))	<b>Si</b> Silicon (28.084, 28.086)	<b>P</b> Phosphorus (30.973761998(5))	<b>S</b> Sulfur (32.059, 32.076)	<b>Cl</b> Chlorine (35.446, 35.457)	<b>Ar</b> Argon (39.792, 39.963)									
4	<b>K</b> Potassium (39.0983(1))	<b>Ca</b> Calcium (40.078(4))	<b>Sc</b> Scandium (44.955908(5))	<b>Ti</b> Titanium (47.867(1))	<b>V</b> Vanadium (50.9415(5))	<b>Cr</b> Chromium (51.9961(6))	<b>Mn</b> Manganese (54.938043(2))	<b>Fe</b> Iron (55.845(2))	<b>Co</b> Cobalt (58.933194(3))	<b>Ni</b> Nickel (58.6934(4))	<b>Cu</b> Copper (63.546(3))	<b>Zn</b> Zinc (65.38(2))	<b>Ga</b> Gallium (69.723(1))	<b>Ge</b> Germanium (72.630(8))	<b>As</b> Arsenic (74.921595(6))	<b>Se</b> Selenium (78.971(8))	<b>Br</b> Bromine (79.901, 79.907)	<b>Kr</b> Krypton (83.798(2))									
5	<b>Rb</b> Rubidium (85.4678(3))	<b>Sr</b> Strontium (87.62(1))	<b>Y</b> Yttrium (88.90584(1))	<b>Zr</b> Zirconium (91.224(2))	<b>Nb</b> Niobium (92.90637(1))	<b>Mo</b> Molybdenum (95.95(1))	<b>Tc</b> Technetium (98)	<b>Ru</b> Ruthenium (101.07(2))	<b>Rh</b> Rhodium (102.90549(2))	<b>Pd</b> Palladium (106.42(1))	<b>Ag</b> Silver (107.8682(2))	<b>Cd</b> Cadmium (112.414(4))	<b>In</b> Indium (114.818(1))	<b>Sn</b> Tin (118.710(7))	<b>Sb</b> Antimony (121.760(1))	<b>Te</b> Tellurium (127.60(3))	<b>I</b> Iodine (126.90447(3))	<b>Xe</b> Xenon (131.293(6))									
6	<b>Cs</b> Caesium (132.90545196(6))	<b>Ba</b> Barium (137.327(7))											<b>Hf</b> Hafnium (178.49(2))	<b>Ta</b> Tantalum (180.94788(2))	<b>W</b> Tungsten (183.84(1))	<b>Re</b> Rhenium (186.207(1))	<b>Os</b> Osmium (190.23(3))	<b>Ir</b> Iridium (192.227(2))	<b>Pt</b> Platinum (195.084(9))	<b>Au</b> Gold (196.966570(4))	<b>Hg</b> Mercury (200.592(3))	<b>Tl</b> Thallium (204.382, 204.385)	<b>Pb</b> Lead (207.2(1))	<b>Bi</b> Bismuth (208.98040(1))	<b>Po</b> Polonium ((209))	<b>At</b> Astatine ((210))	<b>Rn</b> Radon ((222))
7	<b>Fr</b> Francium ((223))	<b>Ra</b> Radium ((226))											<b>Rf</b> Rutherfordium ((261))	<b>Db</b> Dubnium ((268))	<b>Sg</b> Seaborgium ((269))	<b>Bh</b> Bohrium ((271))	<b>Hs</b> Hassium ((277))	<b>Mt</b> Meitnerium ((276))	<b>Ds</b> Darmstadtium ((281))	<b>Rg</b> Roentgenium ((282))	<b>Cn</b> Copernicium ((285))	<b>Nh</b> Nihonium ((286))	<b>Fl</b> Flerovium ((289))	<b>Mc</b> Moscovium ((290))	<b>Lv</b> Livermorium ((293))	<b>Ts</b> Tennessine ((294))	<b>Og</b> Oganesson ((294))
			<b>57</b> <b>La</b> Lanthanum (138.90547(7))	<b>58</b> <b>Ce</b> Cerium (140.116(1))	<b>59</b> <b>Pr</b> Praseodymium (140.90766(1))	<b>60</b> <b>Nd</b> Neodymium (144.242(3))	<b>61</b> <b>Pm</b> Promethium (145)	<b>62</b> <b>Sm</b> Samarium (150.36(2))	<b>63</b> <b>Eu</b> Europium (151.964(1))	<b>64</b> <b>Gd</b> Gadolinium (157.25(3))	<b>65</b> <b>Tb</b> Terbium (158.925354(8))	<b>66</b> <b>Dy</b> Dysprosium (162.500(1))	<b>67</b> <b>Ho</b> Holmium (164.930328(7))	<b>68</b> <b>Er</b> Erbium (167.259(3))	<b>69</b> <b>Tm</b> Thulium (168.934218(6))	<b>70</b> <b>Yb</b> Ytterbium (173.045(10))	<b>71</b> <b>Lu</b> Lutetium (174.9668(1))										
			<b>89</b> <b>Ac</b> Actinium ((227))	<b>90</b> <b>Th</b> Thorium (232.0377(4))	<b>91</b> <b>Pa</b> Protactinium ((231))	<b>92</b> <b>U</b> Uranium (238.02891(3))	<b>93</b> <b>Np</b> Neptunium ((237))	<b>94</b> <b>Pu</b> Plutonium ((244))	<b>95</b> <b>Am</b> Americium ((243))	<b>96</b> <b>Cm</b> Curium ((247))	<b>97</b> <b>Bk</b> Berkelium ((247))	<b>98</b> <b>Cf</b> Californium ((251))	<b>99</b> <b>Es</b> Einsteinium ((252))	<b>100</b> <b>Fm</b> Fermium ((257))	<b>101</b> <b>Md</b> Mendelevium ((258))	<b>102</b> <b>No</b> Nobelium ((259))	<b>103</b> <b>Lr</b> Lawrencium ((262))										

**Atomic number** → 6

**Symbol** → C

**Name** → Carbon

**Ground-state Configuration** → 1s<sup>2</sup>2s<sup>2</sup>2p<sup>2</sup>

**Ground-state level** → 3p<sub>0</sub>

**Standard Atomic Weight<sup>1</sup>** → 12.011<sup>1</sup>

**Ionization Energy (eV)<sup>2</sup>** → 11.2603

<sup>1</sup> Based upon <sup>12</sup>C. Reported from CIAAW *Atomic Weights of the Elements 2017* (ciaaw.org) and updated to reflect the 05 June 2018 IUPAC revised atomic weights. Brackets indicate the range of values in which the atomic weight of a terrestrial element is expected to fall. Parentheses indicate the range (+/-) in which the atomic weight is expected to fall in normal materials. For example, the weight of iridium, 192.217(2), is expected to be between 192.215 and 192.219.

<sup>2</sup> Reported values from NIST, 2019.

**Atomic weight exceptions:**  
(O) Mass number of longest lived isotope reported.

- Solids
- Liquids
- Gases
- Artificially Prepared
- Radioactive

Group		1 IA										18 VIIIA									
1	<sup>1</sup> s <sub>1/2</sub>	2 IIA										2									
1	<b>H</b> Hydrogen (1.00784, 1.00811)	3 IIA										2									
2	<sup>1</sup> s <sup>2</sup>	4 IIA										2									
2	<b>Li</b> Lithium (6.938, 6.997)	<b>Be</b> Beryllium (9.0121831(5))										3 IIA									
3	<sup>1</sup> s <sup>2</sup> <sup>2</sup> s <sup>1</sup>	<sup>1</sup> s <sup>2</sup> <sup>2</sup> s <sup>2</sup>										3 IIA									
3	<b>Na</b> Sodium (22.98976928(2))	<b>Mg</b> Magnesium (24.304, 24.307)										3 IIA									
3	[Ne]3s <sup>1</sup> 5.1391	[Ne]3s <sup>2</sup> 7.6462										3 IIA									
4	<sup>1</sup> s <sup>2</sup> <sup>2</sup> s <sup>2</sup> <sup>2</sup> p <sup>1</sup>	3 IIIB										3 IIA									
4	<b>K</b> Potassium (39.0983(1))	<b>Ca</b> Calcium (40.078(4))										3 IIA									
4	[Ar]4s <sup>1</sup> 4.3407	[Ar]4s <sup>2</sup> 6.1132										3 IIA									
5	<sup>1</sup> s <sup>2</sup> <sup>2</sup> s <sup>2</sup> <sup>2</sup> p <sup>6</sup> <sup>3</sup> d <sup>1</sup>	3 IIIB										3 IIA									
5	<b>Rb</b> Rubidium (85.4678(3))	<b>Sr</b> Strontium (87.62(1))										3 IIA									
5	[Kr]5s <sup>1</sup> 4.7171	[Kr]5s <sup>2</sup> 5.6949										3 IIA									
6	<sup>1</sup> s <sup>2</sup> <sup>2</sup> s <sup>2</sup> <sup>2</sup> p <sup>6</sup> <sup>3</sup> d <sup>10</sup> <sup>4</sup> s <sup>1</sup>	3 IIIB										3 IIA									
6	<b>Cs</b> Caesium (132.90545196(6))	<b>Ba</b> Barium (137.327(7))										3 IIA									
6	[Xe]6s <sup>1</sup> 3.8939	[Xe]6s <sup>2</sup> 5.2117										3 IIA									
7	<sup>1</sup> s <sup>2</sup> <sup>2</sup> s <sup>2</sup> <sup>2</sup> p <sup>6</sup> <sup>3</sup> d <sup>10</sup> <sup>4</sup> s <sup>2</sup> <sup>4</sup> p <sup>1</sup>	3 IIIB										3 IIA									
7	<b>Fr</b> Francium ([223])	<b>Ra</b> Radium ([226])										3 IIA									
7	[Rn]7s <sup>1</sup> 4.0727	[Rn]7s <sup>2</sup> 5.704										3 IIA									

**Carbon**

Atomic number: 6

Symbol: C

Name: Carbon

Ground-state Configuration: 1s<sup>2</sup>2s<sup>2</sup>2p<sup>2</sup>

Ground-state level: <sup>3</sup>P<sub>0</sub>

Standard Atomic Weight: 12.011

Ionization Energy (eV)<sup>2</sup>: 11.2603

<sup>1</sup> Based upon <sup>12</sup>C. Reported from CIAAW *Atomic Weights of the Elements 2017* (ciaaw.org) and updated to reflect the 05 June 2018 IUPAC revised atomic weights. Brackets indicate the range of values in which the atomic weight of a terrestrial element is expected to fall. Parentheses indicate the range (1/2) in which the atomic weight is expected to fall in normal materials. For example, the weight of Iridium, 192.217(2), is expected to be between 192.215 and 192.219.

<sup>2</sup> Reported values from NIST, 2019.

Atomic weight exceptions:  
(<sup>1</sup>) Mass number of longest lived isotope reported.

**Legend:**

- Solids (Light Blue)
- Liquids (Dark Blue)
- Gases (Green)
- Artificially Prepared (Grey)
- Radioactive (Red)

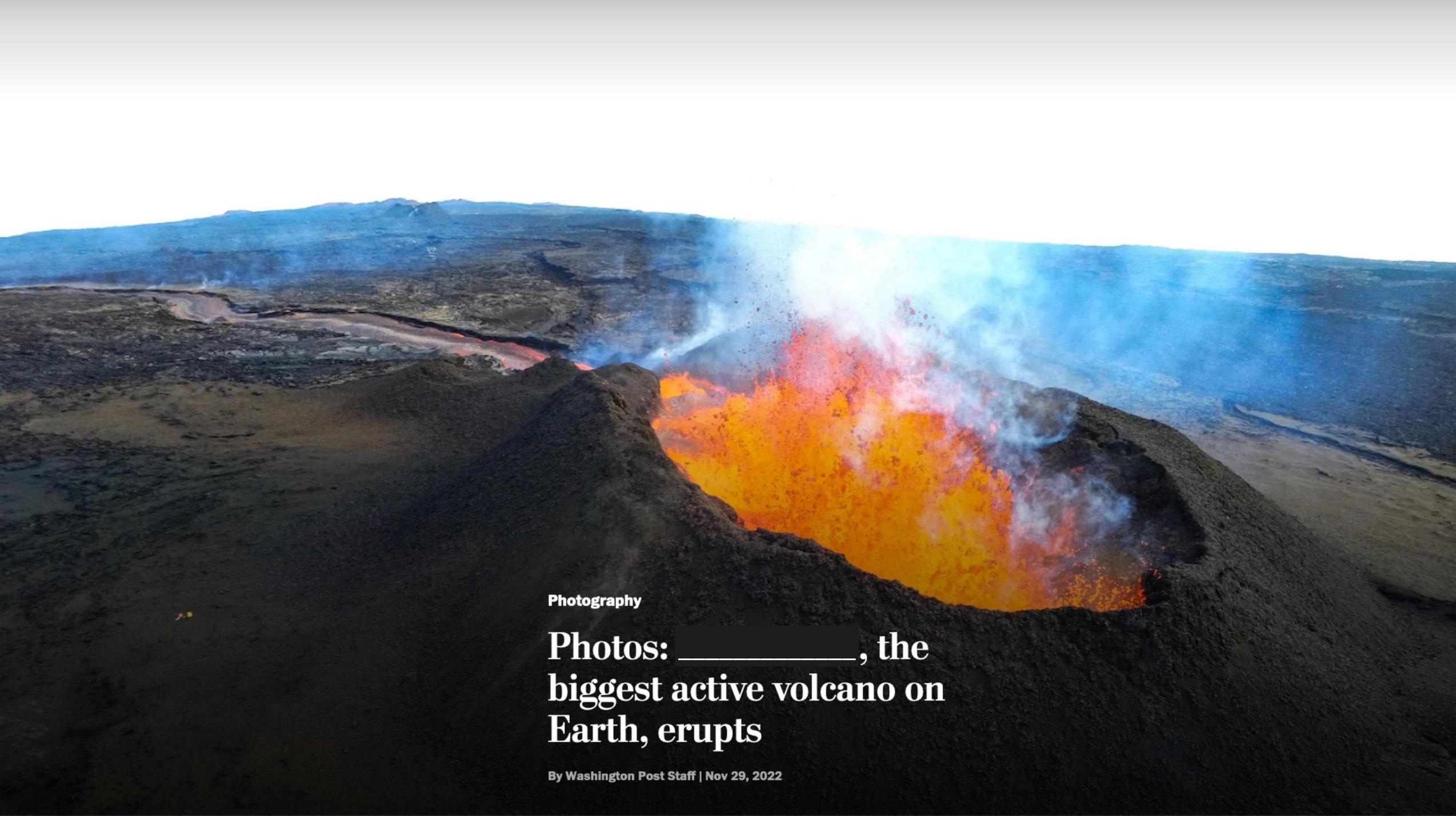
LANTHANIDES	57	<sup>4</sup> f <sup>7</sup> <sup>5</sup> d <sup>0</sup> <sup>6</sup> s <sup>2</sup>	<b>La</b> Lanthanum (138.90547(7))	58	<sup>4</sup> f <sup>0</sup> <sup>5</sup> d <sup>1</sup> <sup>6</sup> s <sup>2</sup>	<b>Ce</b> Cerium (140.116(1))	59	<sup>4</sup> f <sup>1</sup> <sup>5</sup> d <sup>0</sup> <sup>6</sup> s <sup>2</sup>	<b>Pr</b> Praseodymium (140.90766(1))	60	<sup>4</sup> f <sup>2</sup> <sup>5</sup> d <sup>0</sup> <sup>6</sup> s <sup>2</sup>	<b>Nd</b> Neodymium (144.242(3))	61	<sup>4</sup> f <sup>3</sup> <sup>5</sup> d <sup>0</sup> <sup>6</sup> s <sup>2</sup>	<b>Pm</b> Promethium ([145])	62	<sup>4</sup> f <sup>4</sup> <sup>5</sup> d <sup>0</sup> <sup>6</sup> s <sup>2</sup>	<b>Sm</b> Samarium (150.36(2))	63	<sup>4</sup> f <sup>5</sup> <sup>5</sup> d <sup>0</sup> <sup>6</sup> s <sup>2</sup>	<b>Eu</b> Europium (151.964(1))	64	<sup>4</sup> f <sup>6</sup> <sup>5</sup> d <sup>0</sup> <sup>6</sup> s <sup>2</sup>	<b>Gd</b> Gadolinium (157.25(3))	65	<sup>4</sup> f <sup>7</sup> <sup>5</sup> d <sup>0</sup> <sup>6</sup> s <sup>2</sup>	<b>Tb</b> Terbium (158.925354(8))	66	<sup>4</sup> f <sup>7</sup> <sup>5</sup> d <sup>1</sup> <sup>6</sup> s <sup>2</sup>	<b>Dy</b> Dysprosium (162.500(1))	67	<sup>4</sup> f <sup>9</sup> <sup>5</sup> d <sup>0</sup> <sup>6</sup> s <sup>2</sup>	<b>Ho</b> Holmium (164.930328(7))	68	<sup>4</sup> f <sup>10</sup> <sup>5</sup> d <sup>0</sup> <sup>6</sup> s <sup>2</sup>	<b>Er</b> Erbium (167.259(3))	69	<sup>4</sup> f <sup>11</sup> <sup>5</sup> d <sup>0</sup> <sup>6</sup> s <sup>2</sup>	<b>Tm</b> Thulium (168.934218(6))	70	<sup>4</sup> f <sup>12</sup> <sup>5</sup> d <sup>0</sup> <sup>6</sup> s <sup>2</sup>	<b>Yb</b> Ytterbium (173.045(1))	71	<sup>4</sup> f <sup>13</sup> <sup>5</sup> d <sup>0</sup> <sup>6</sup> s <sup>2</sup>	<b>Lu</b> Lutetium (174.9668(1))
	ACTINIDES	89	<sup>5</sup> f <sup>7</sup> <sup>6</sup> d <sup>0</sup> <sup>7</sup> s <sup>2</sup>	<b>Ac</b> Actinium ([227])	90	<sup>5</sup> f <sup>0</sup> <sup>6</sup> d <sup>1</sup> <sup>7</sup> s <sup>2</sup>	<b>Th</b> Thorium (232.0377(4))	91	<sup>5</sup> f <sup>1</sup> <sup>6</sup> d <sup>0</sup> <sup>7</sup> s <sup>2</sup>	<b>Pa</b> Protactinium ([231])	92	<sup>5</sup> f <sup>2</sup> <sup>6</sup> d <sup>0</sup> <sup>7</sup> s <sup>2</sup>	<b>U</b> Uranium (238.02891(3))	93	<sup>5</sup> f <sup>3</sup> <sup>6</sup> d <sup>0</sup> <sup>7</sup> s <sup>2</sup>	<b>Np</b> Neptunium ([237])	94	<sup>5</sup> f <sup>4</sup> <sup>6</sup> d <sup>0</sup> <sup>7</sup> s <sup>2</sup>	<b>Pu</b> Plutonium ([244])	95	<sup>5</sup> f <sup>5</sup> <sup>6</sup> d <sup>0</sup> <sup>7</sup> s <sup>2</sup>	<b>Am</b> Americium ([243])	96	<sup>5</sup> f <sup>6</sup> <sup>6</sup> d <sup>0</sup> <sup>7</sup> s <sup>2</sup>	<b>Cm</b> Curium ([247])	97	<sup>5</sup> f <sup>7</sup> <sup>6</sup> d <sup>0</sup> <sup>7</sup> s <sup>2</sup>	<b>Bk</b> Berkelium ([247])	98	<sup>5</sup> f <sup>7</sup> <sup>6</sup> d <sup>1</sup> <sup>7</sup> s <sup>2</sup>	<b>Cf</b> Californium ([251])	99	<sup>5</sup> f <sup>9</sup> <sup>6</sup> d <sup>0</sup> <sup>7</sup> s <sup>2</sup>	<b>Es</b> Einsteinium ([252])	100	<sup>5</sup> f <sup>10</sup> <sup>6</sup> d <sup>0</sup> <sup>7</sup> s <sup>2</sup>	<b>Fm</b> Fermium ([257])	101	<sup>5</sup> f <sup>11</sup> <sup>6</sup> d <sup>0</sup> <sup>7</sup> s <sup>2</sup>	<b>Md</b> Mendelevium ([258])	102	<sup>5</sup> f <sup>12</sup> <sup>6</sup> d <sup>0</sup> <sup>7</sup> s <sup>2</sup>	<b>No</b> Nobelium ([259])	103	<sup>5</sup> f <sup>13</sup> <sup>6</sup> d <sup>0</sup> <sup>7</sup> s <sup>2</sup>











Photography

# Photos: \_\_\_\_\_, the biggest active volcano on Earth, erupts

By Washington Post Staff | Nov 29, 2022





Photography

# Photos: Mauna Loa, the biggest active volcano on Earth, erupts

By Washington Post Staff | Nov 29, 2022

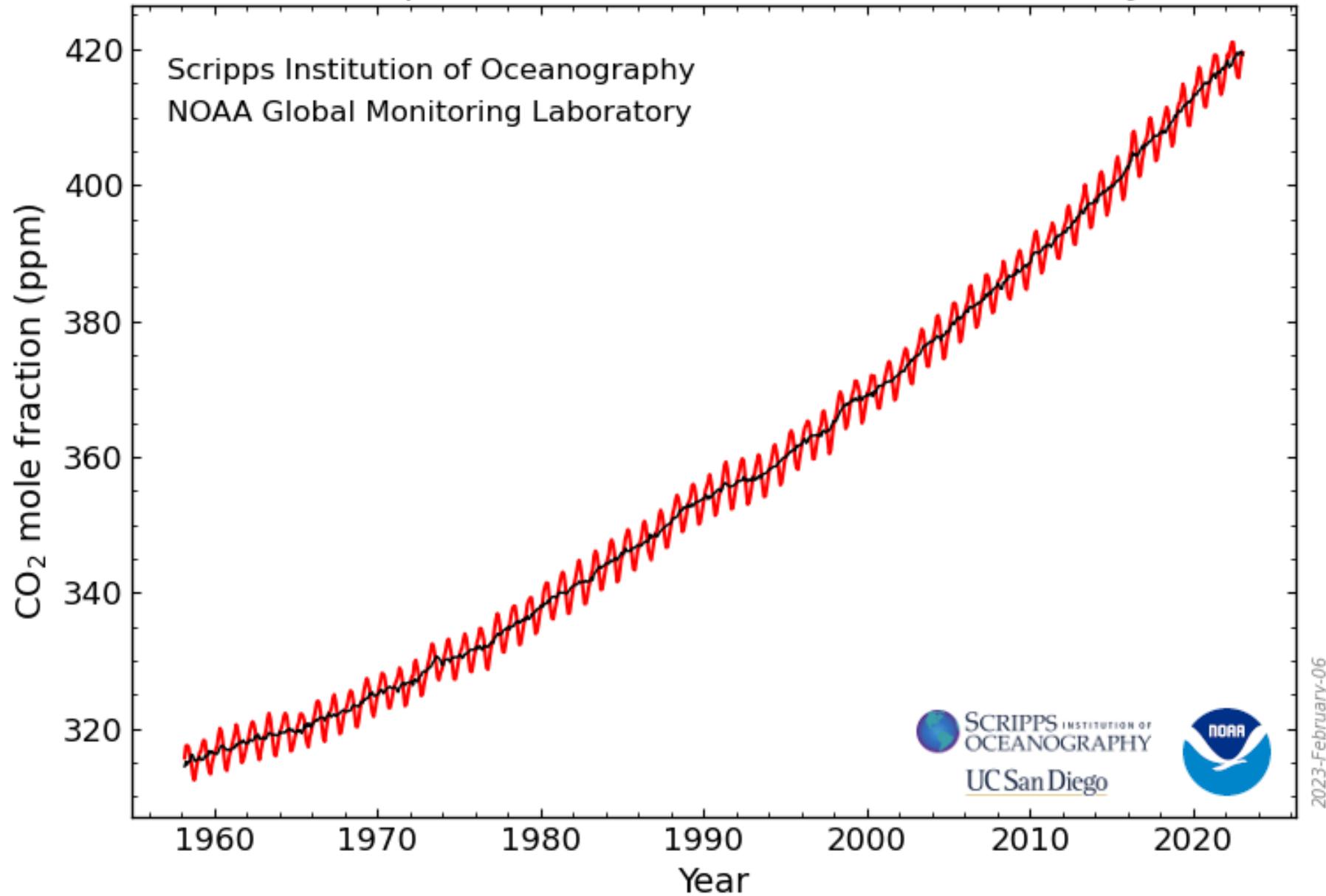


# MAUNA LOA OBSERVATORY

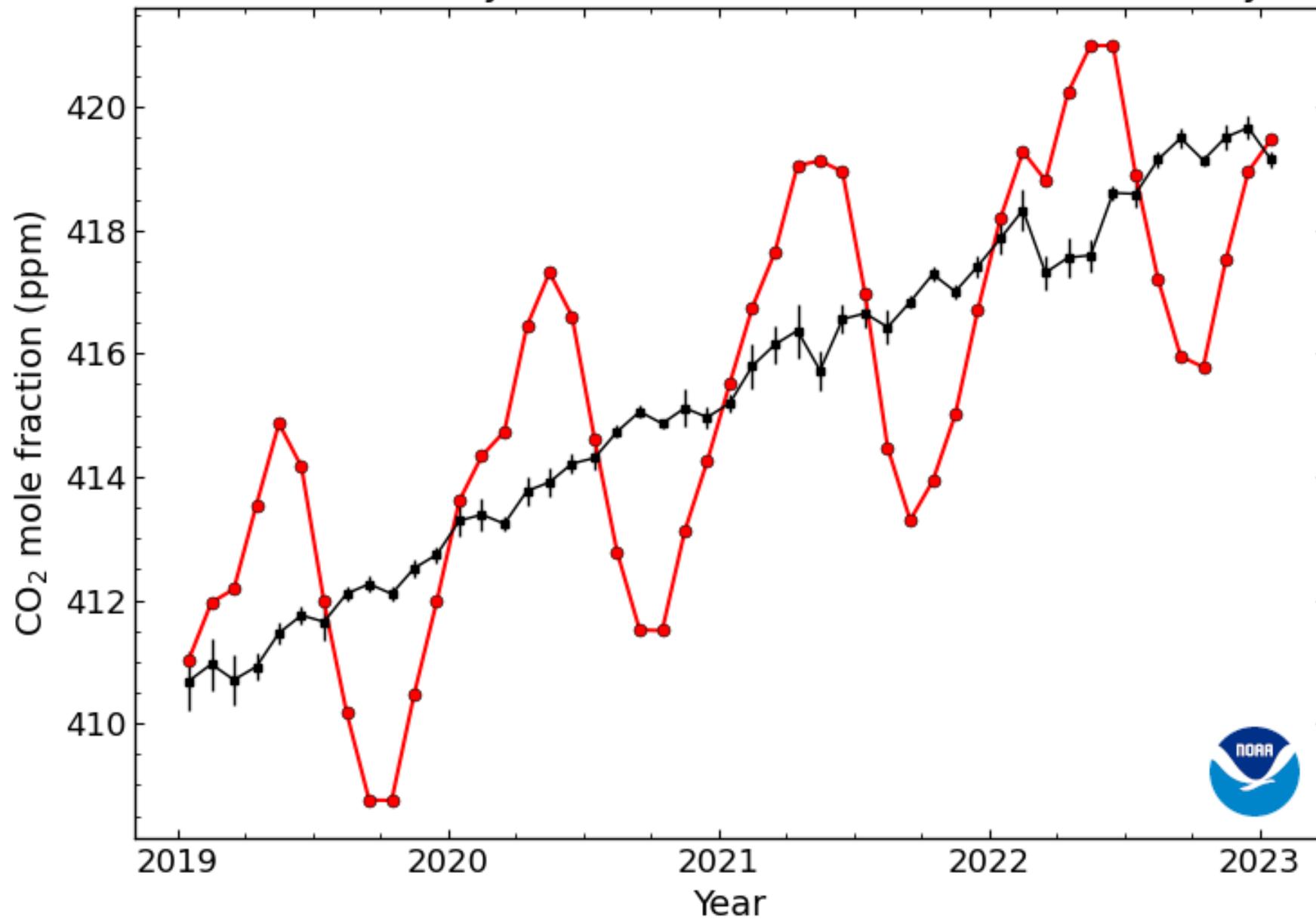
Earth System Research Laboratories  
Global Monitoring Division

Est. June 28, 1956 • Elevation 3,396m (11,141 ft)

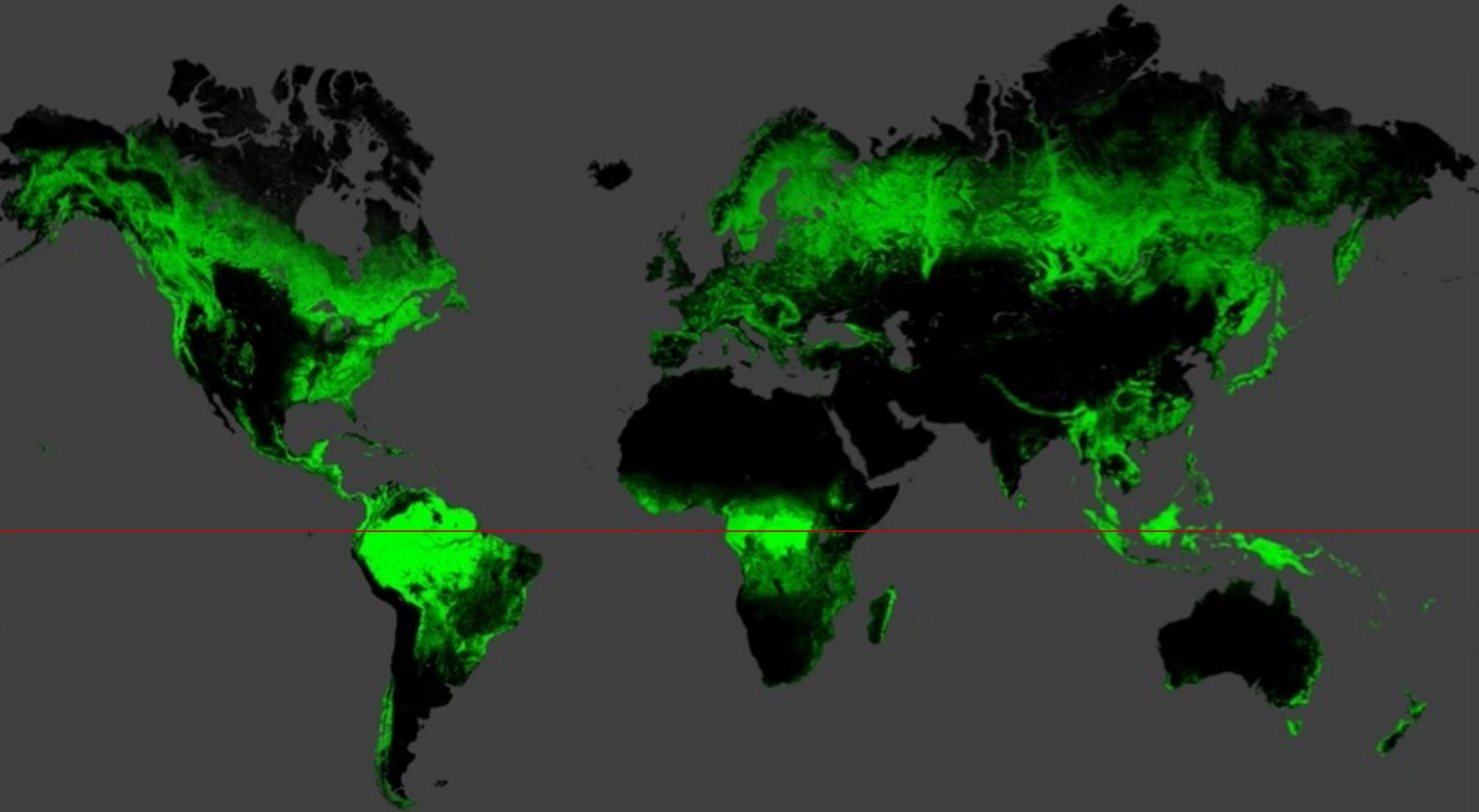
# Atmospheric CO<sub>2</sub> at Mauna Loa Observatory



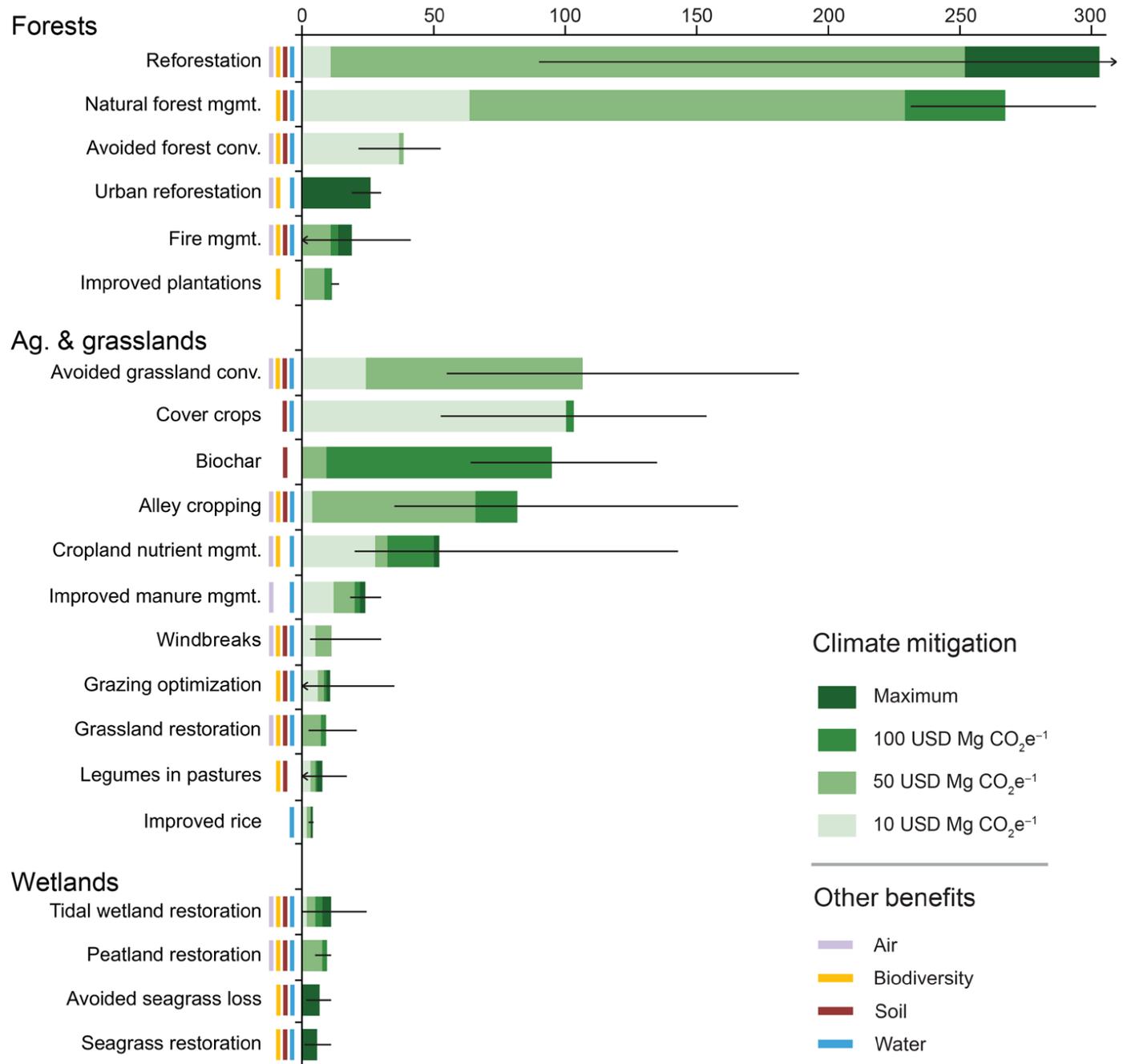
# Recent Monthly Mean CO<sub>2</sub> at Mauna Loa Observatory



2023-February-06

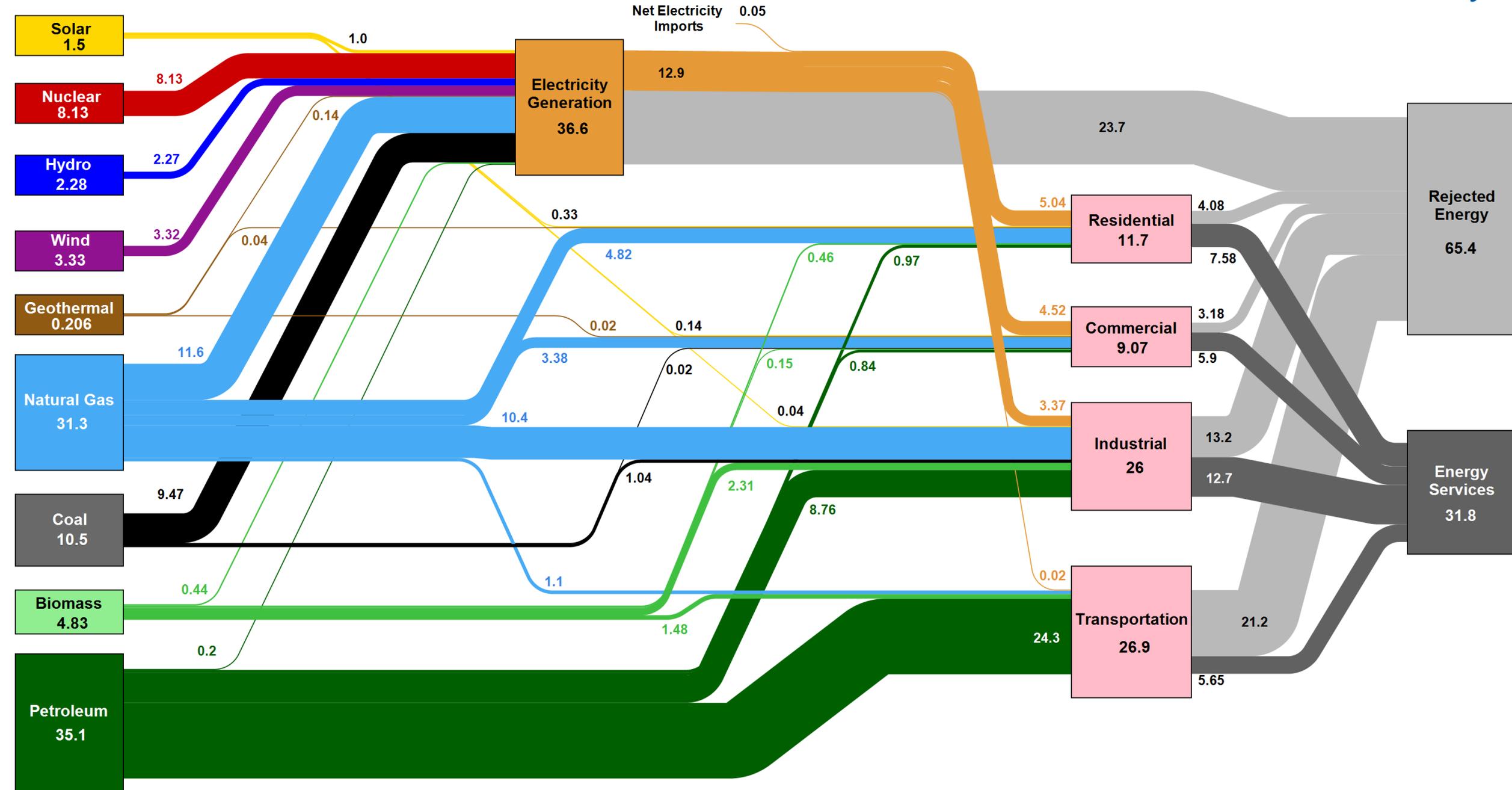


# Climate mitigation potential in 2025 (Tg CO<sub>2</sub>e year<sup>-1</sup>)





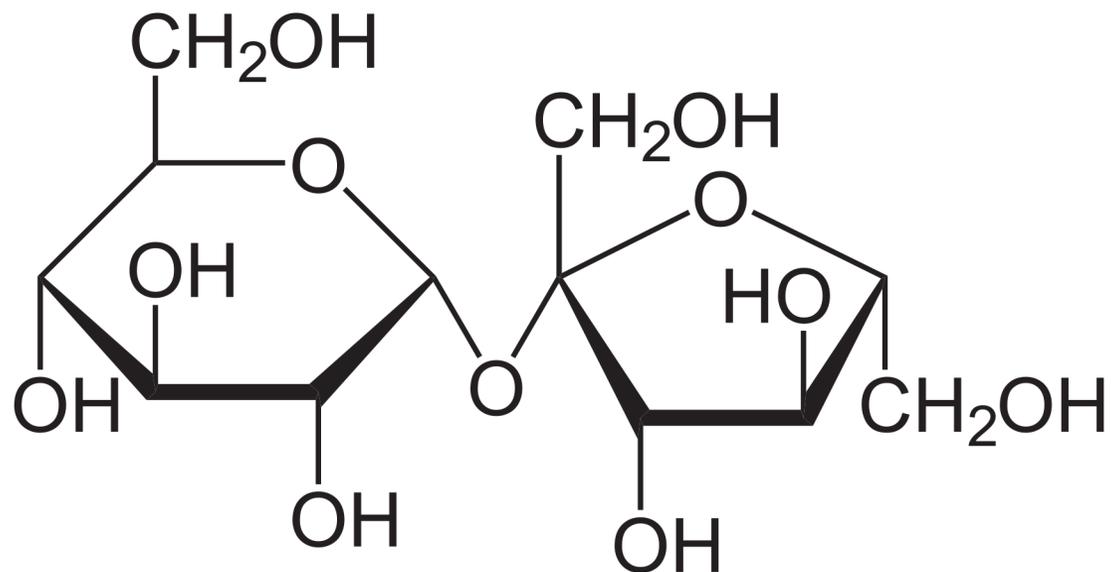
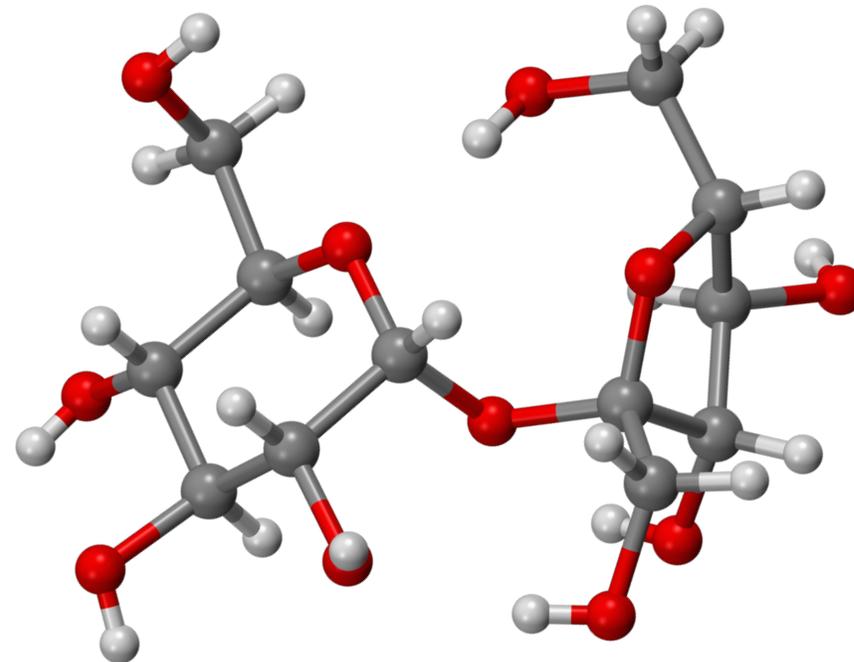
# Estimated U.S. Energy Consumption in 2021: 97.3 Quads



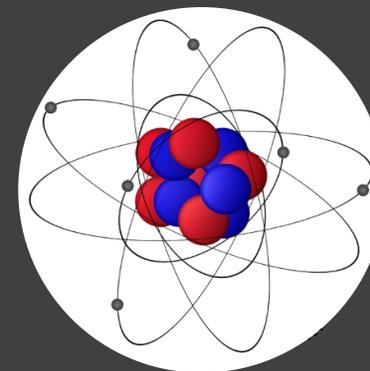


# Carbon Accounting

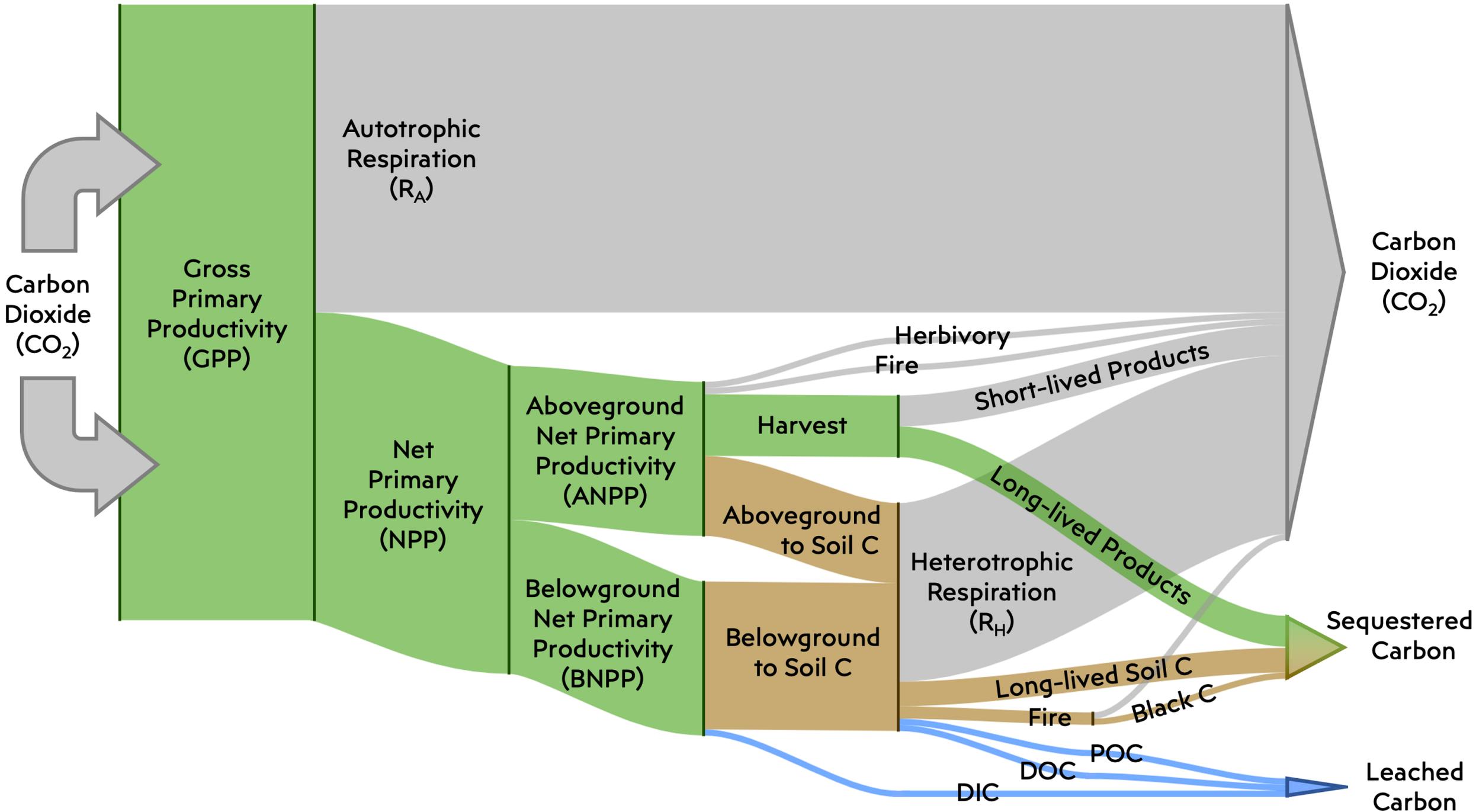
example: sucrose  
(a.k.a. table sugar)

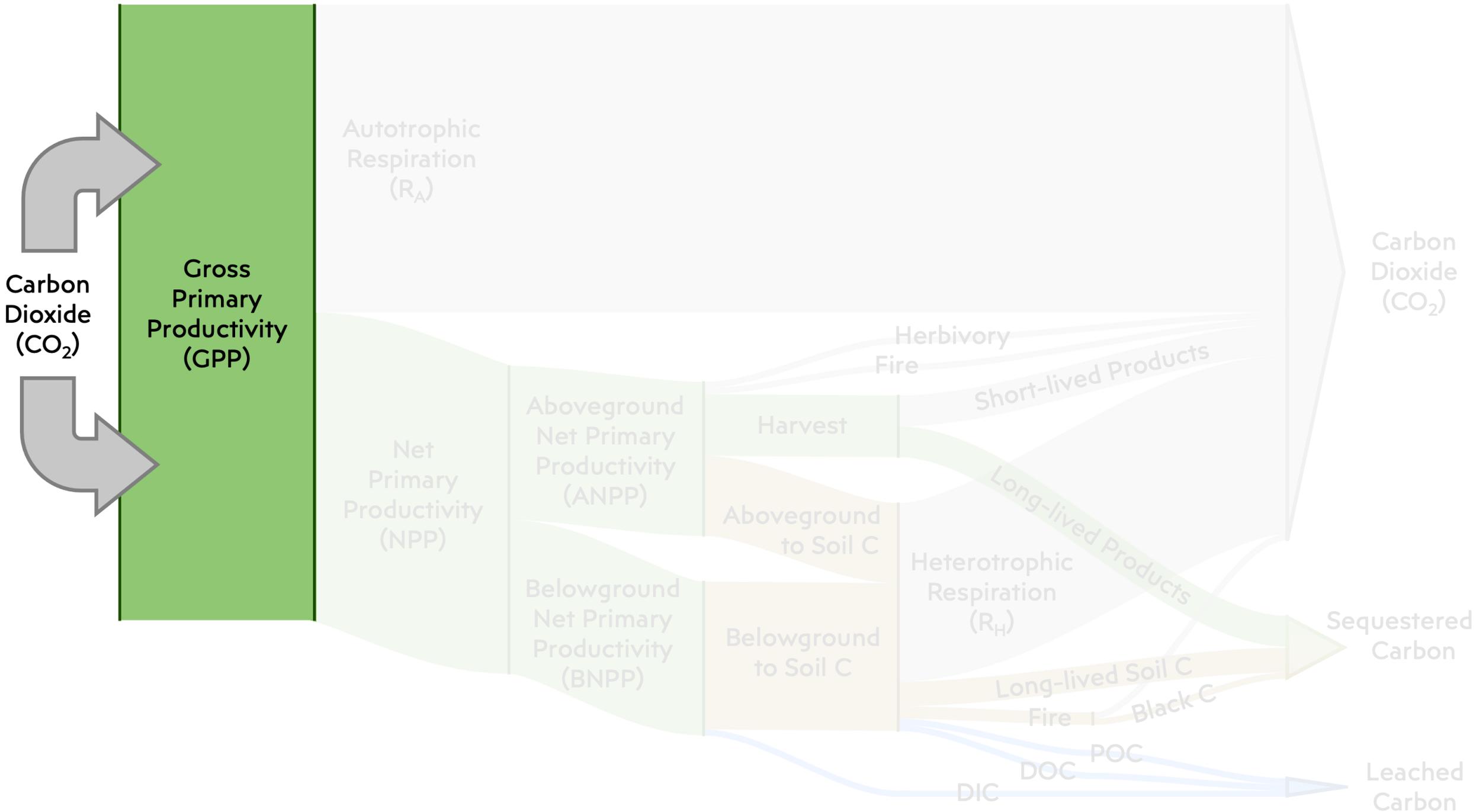


84,000,000,000,000,000,000,000,000  
(or 84 sextillion)







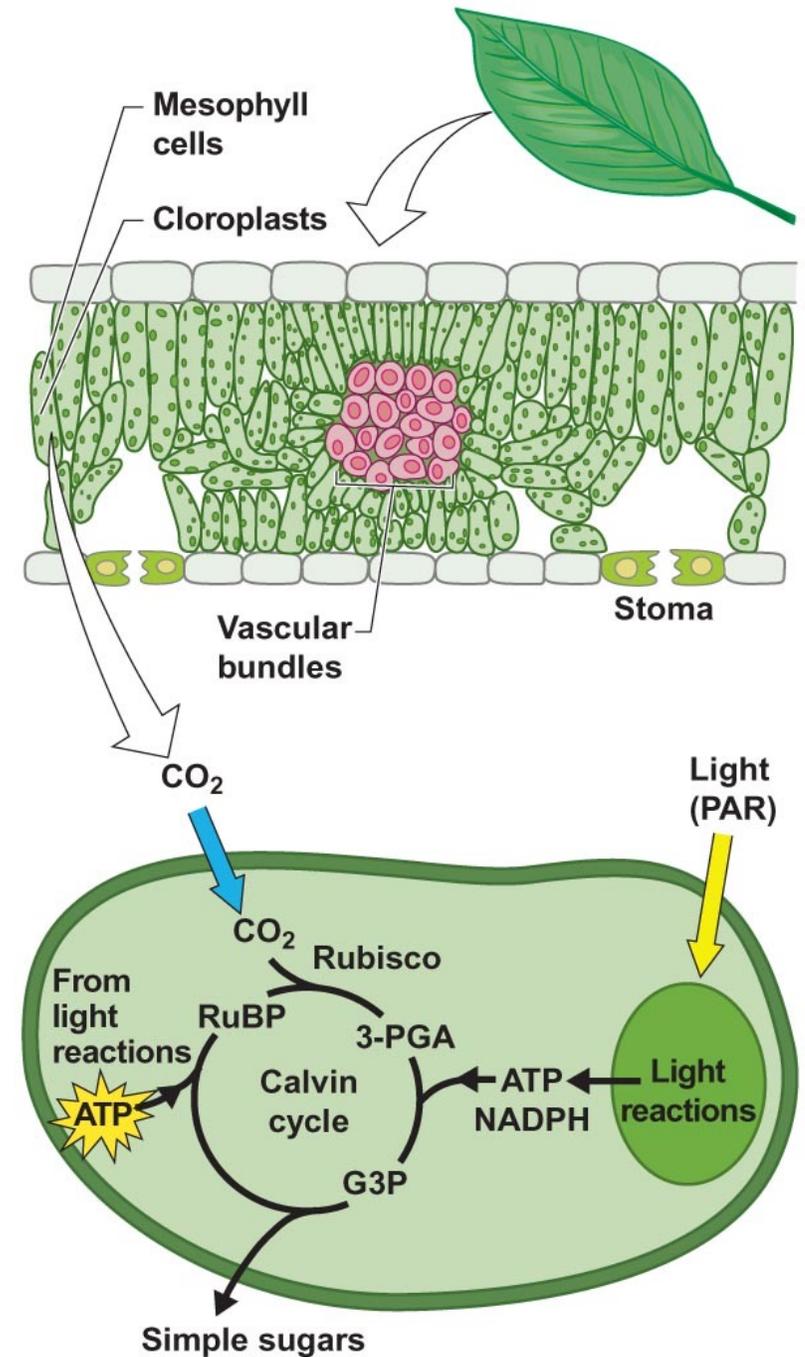


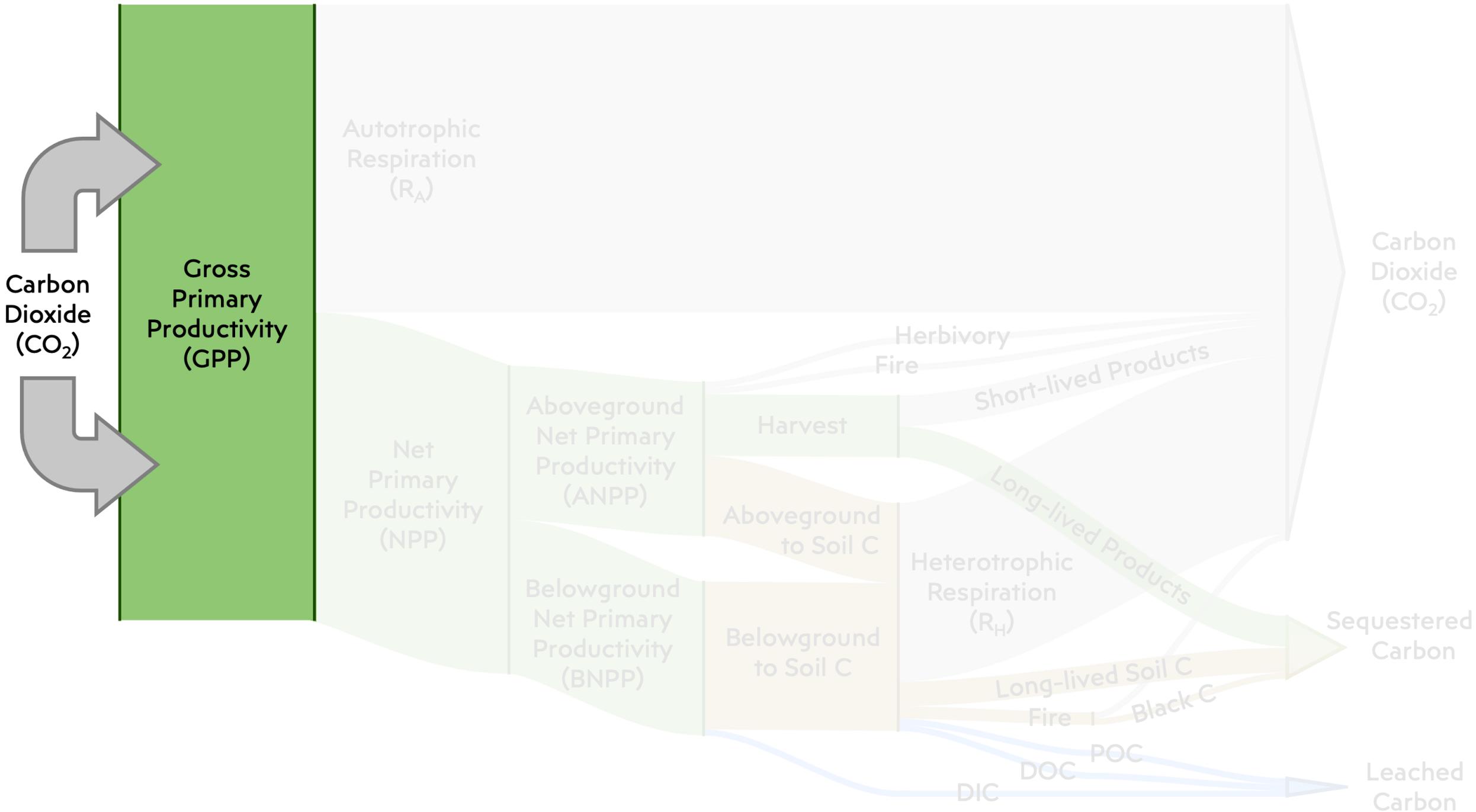
# Photosynthesis

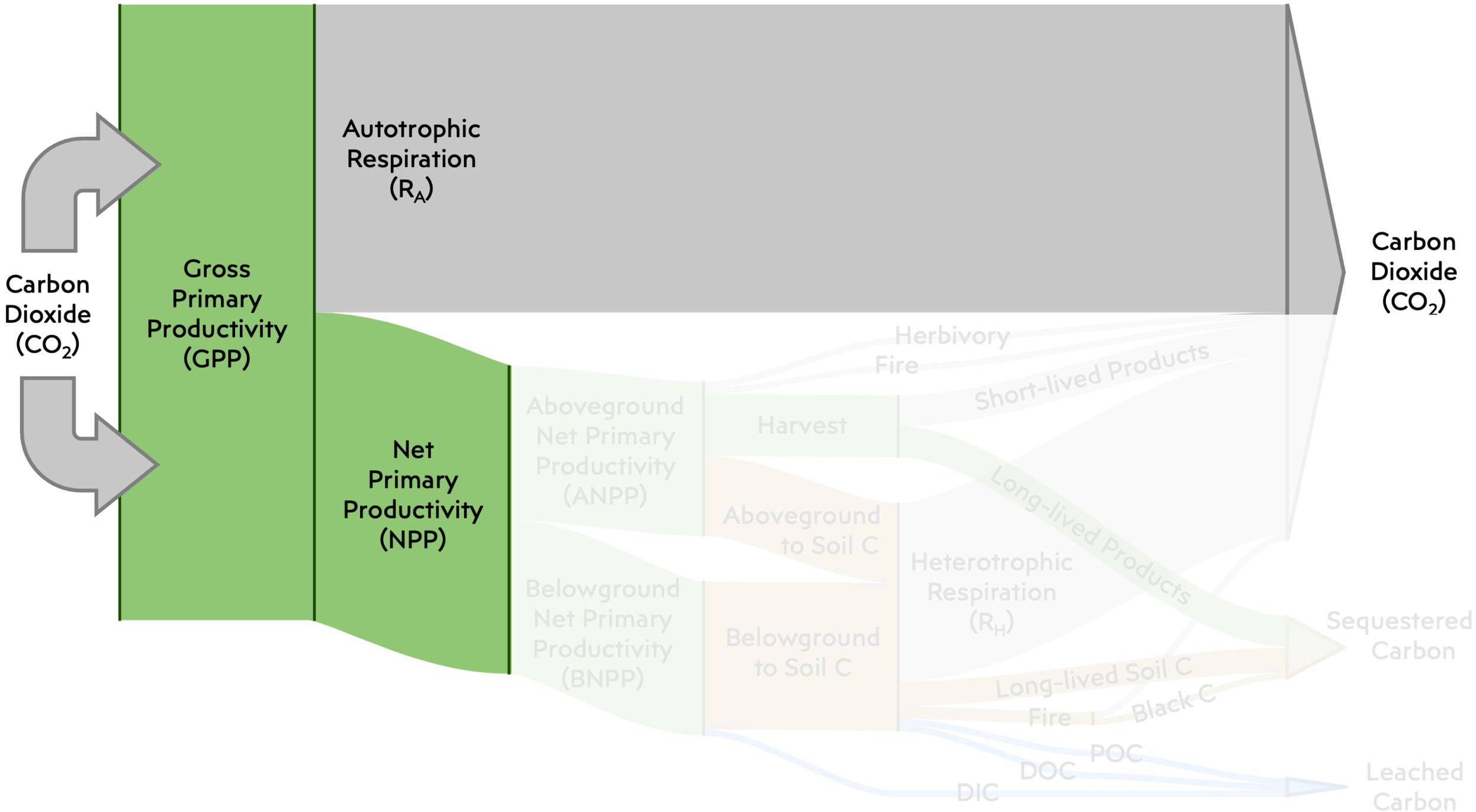


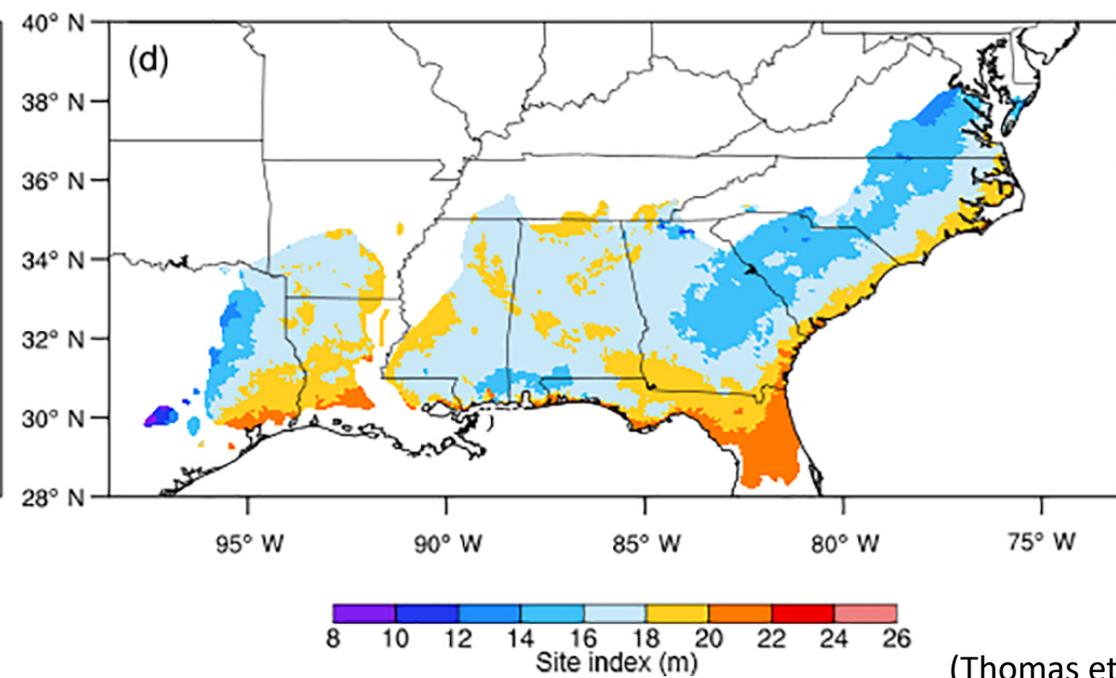
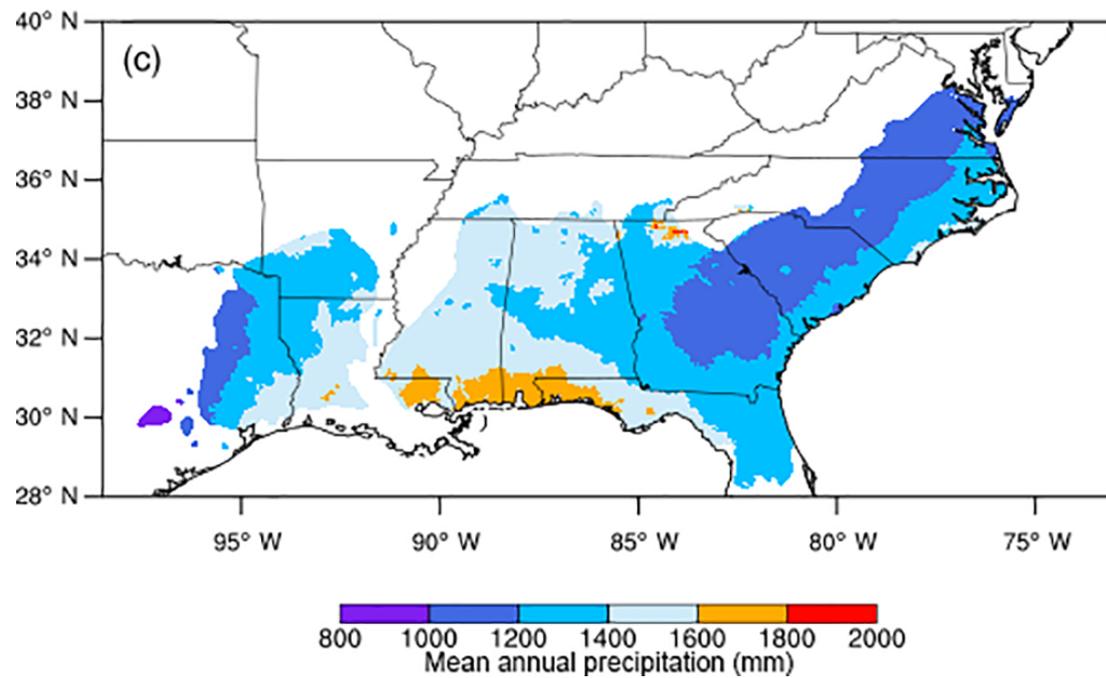
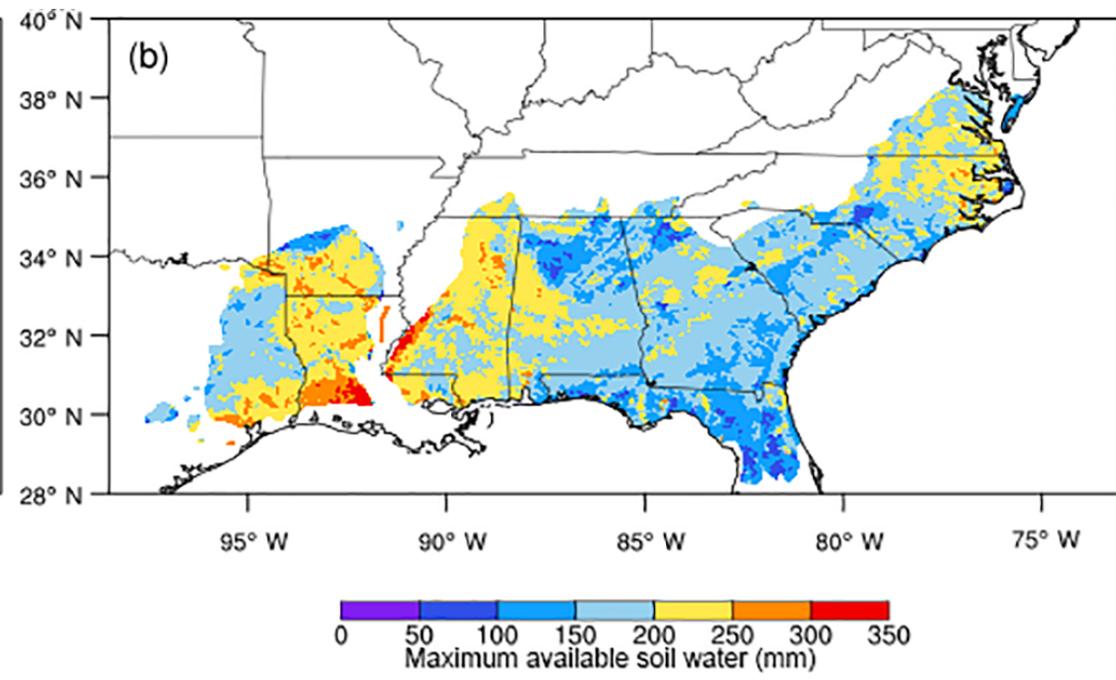
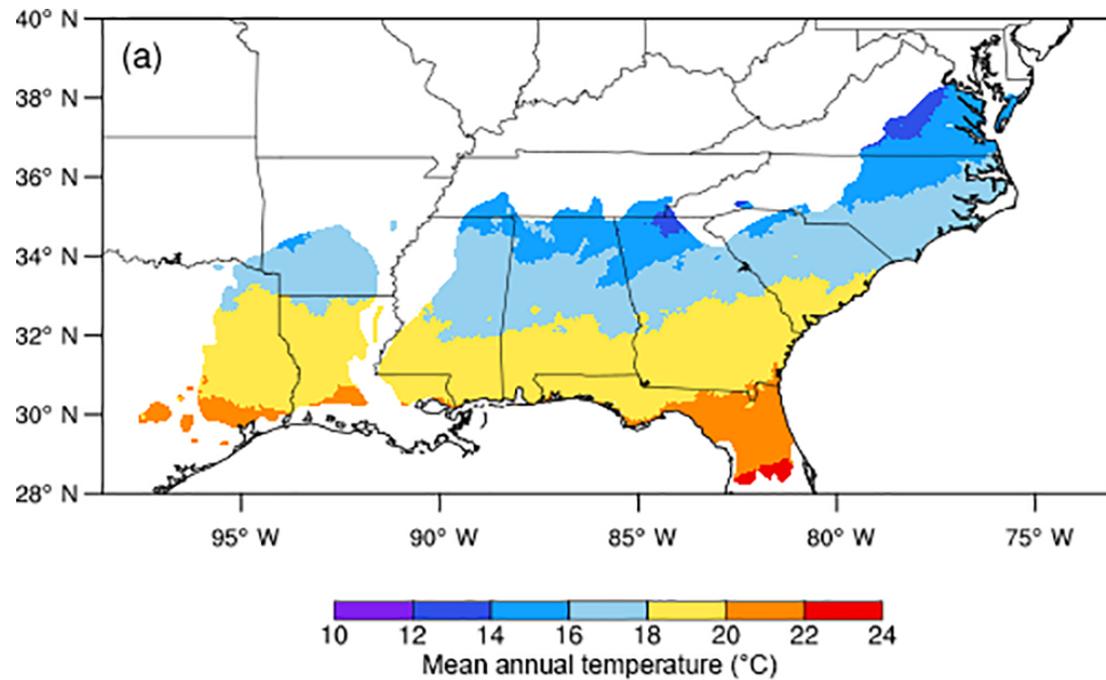
or...

...in the presence of water, sunlight provides the energy to drive the conversion of carbon dioxide into sugar and produces oxygen as a by-product.

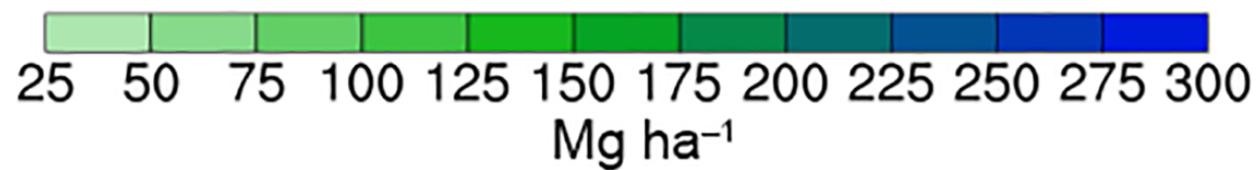
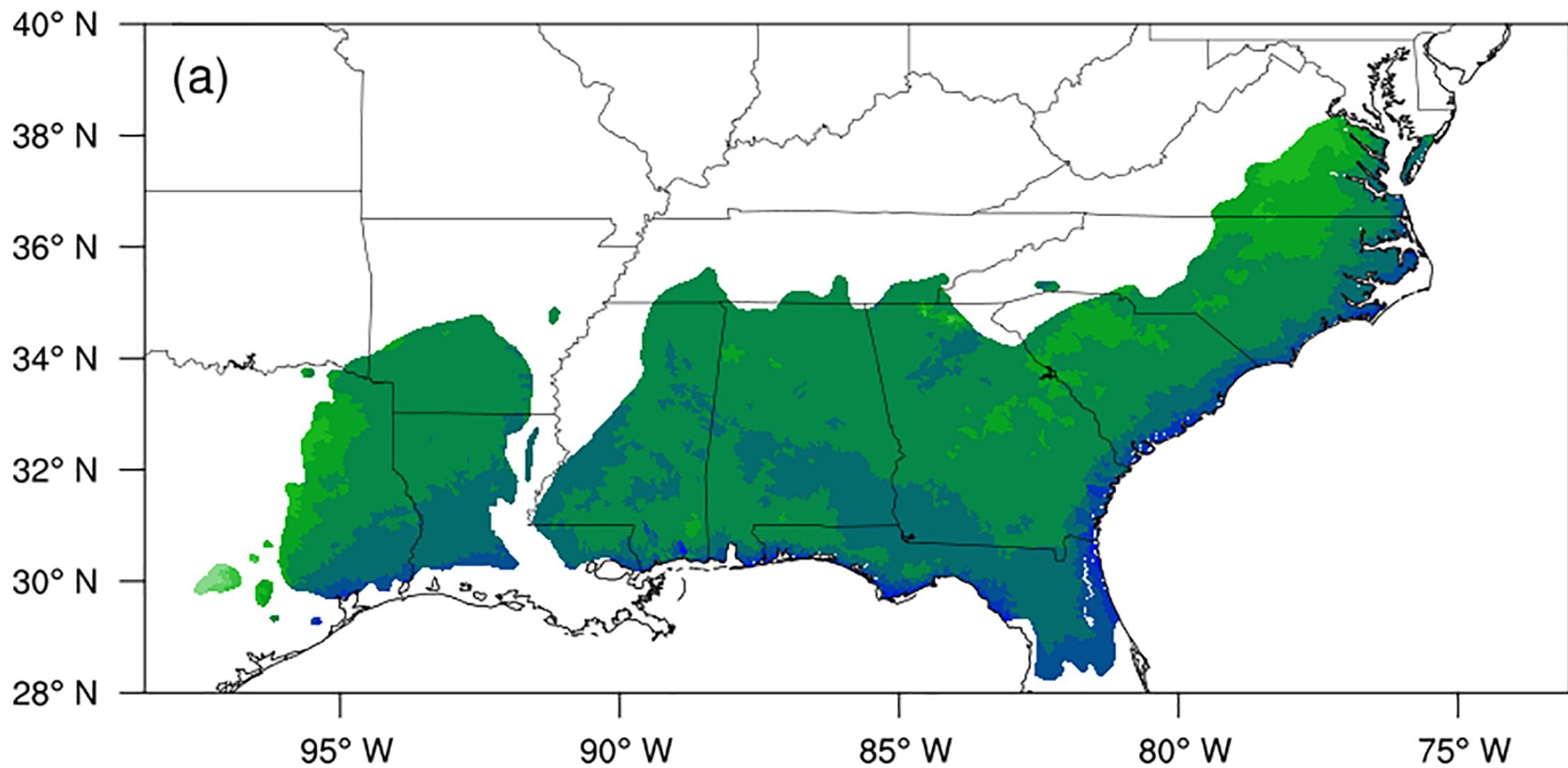


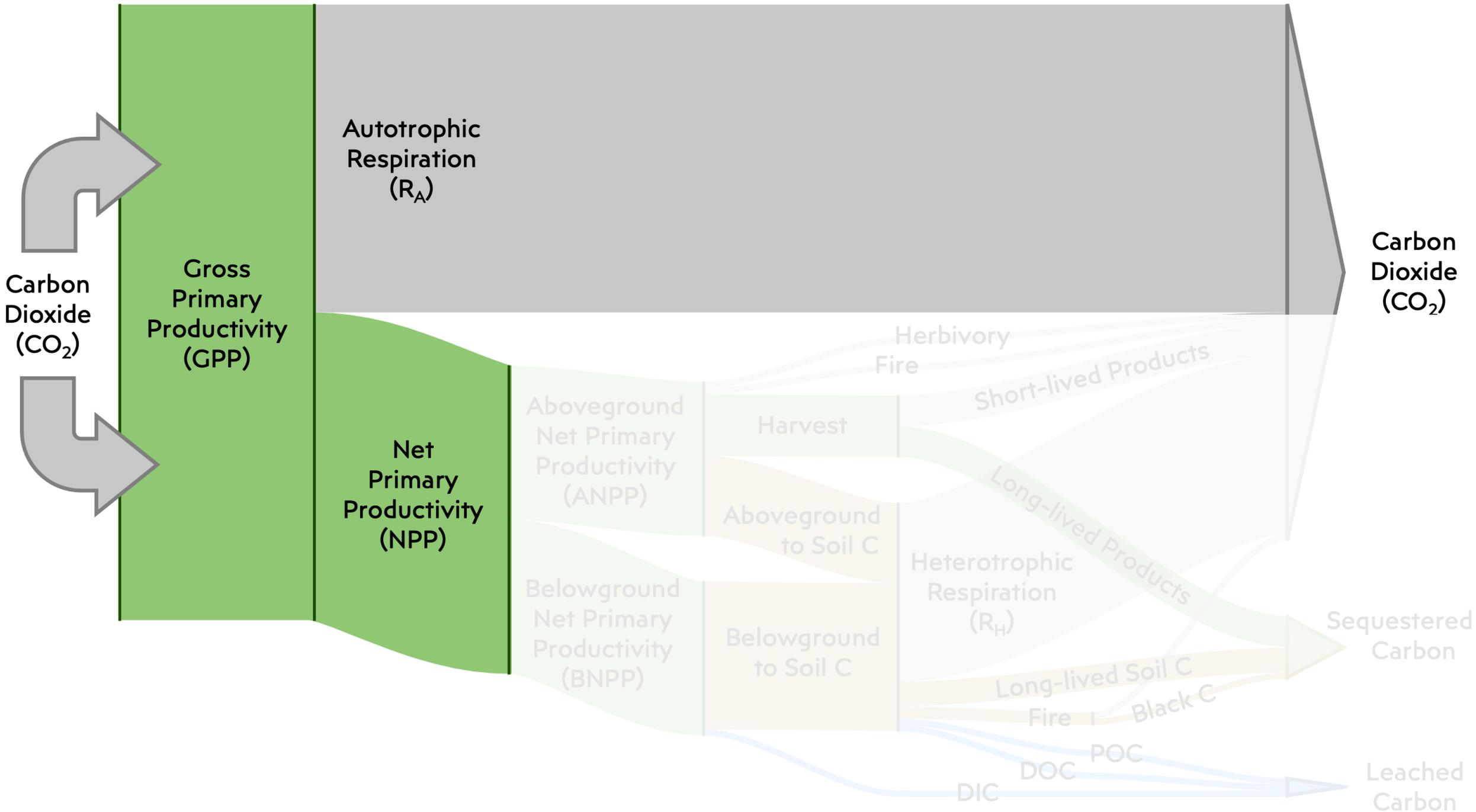


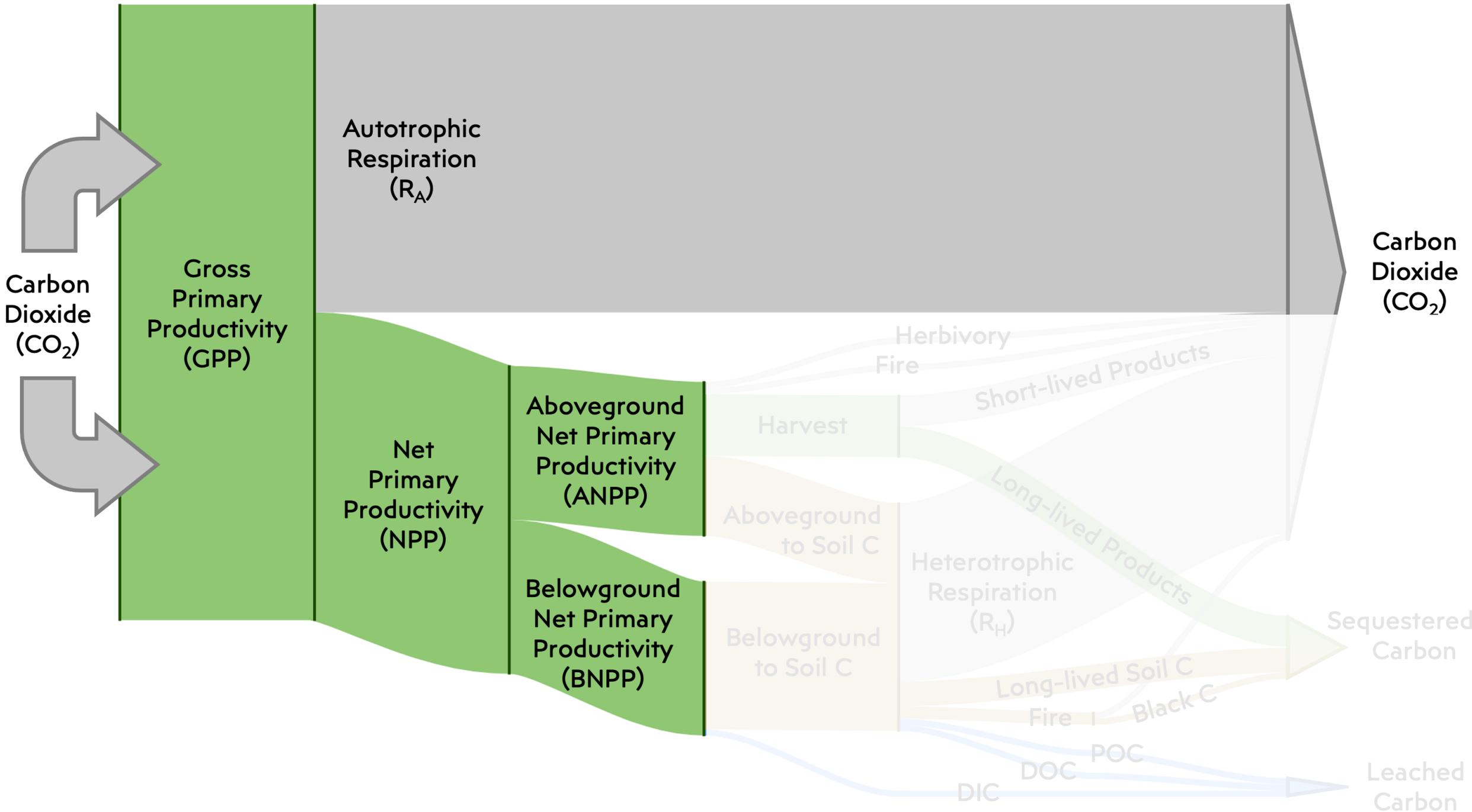








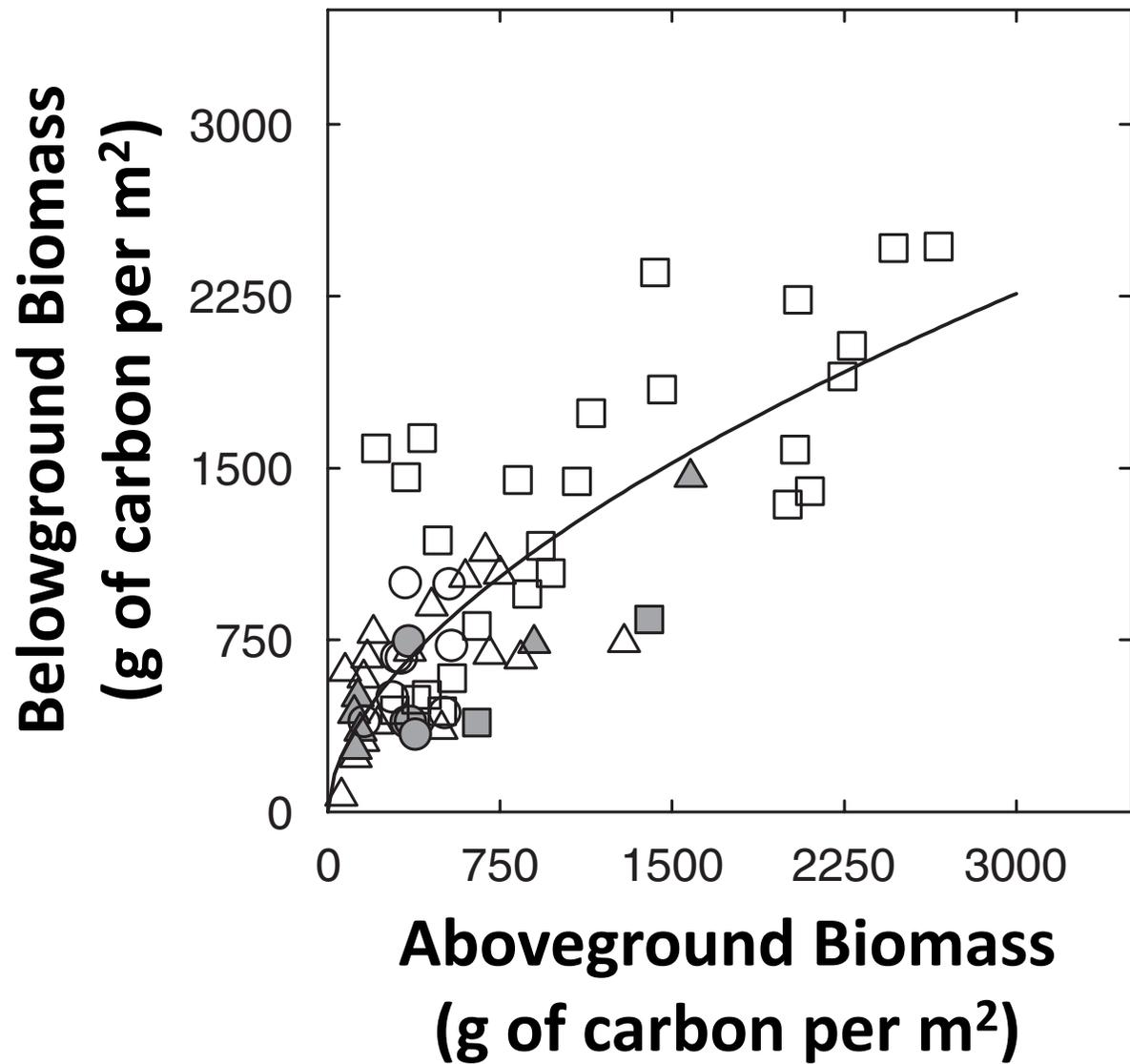




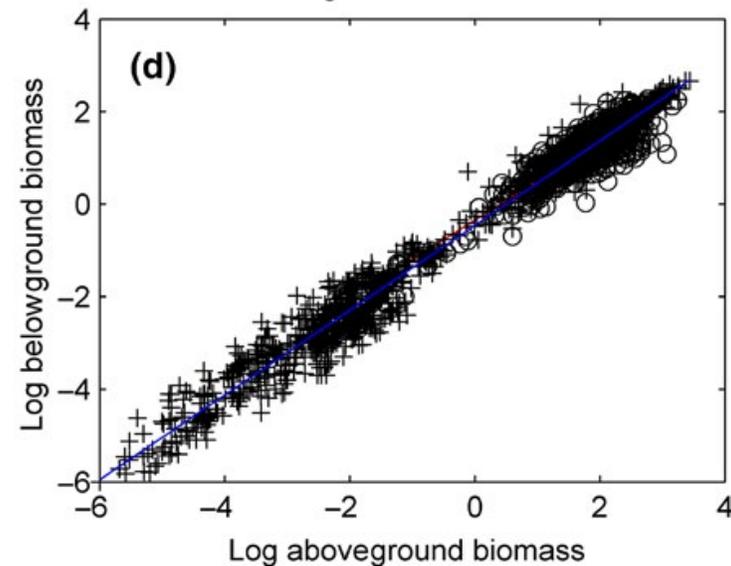
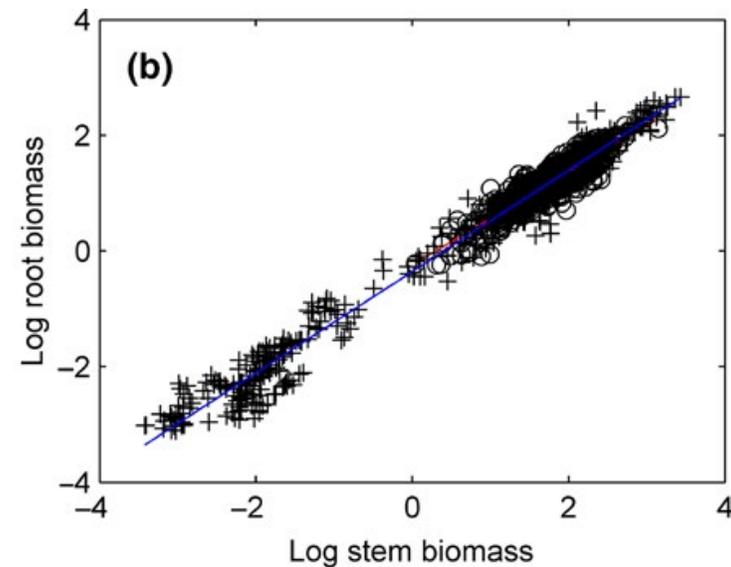


*“The taproot of this longleaf pine stump in Mississippi has a diameter of 14 inches at a depth of 10 feet below the surface.”*

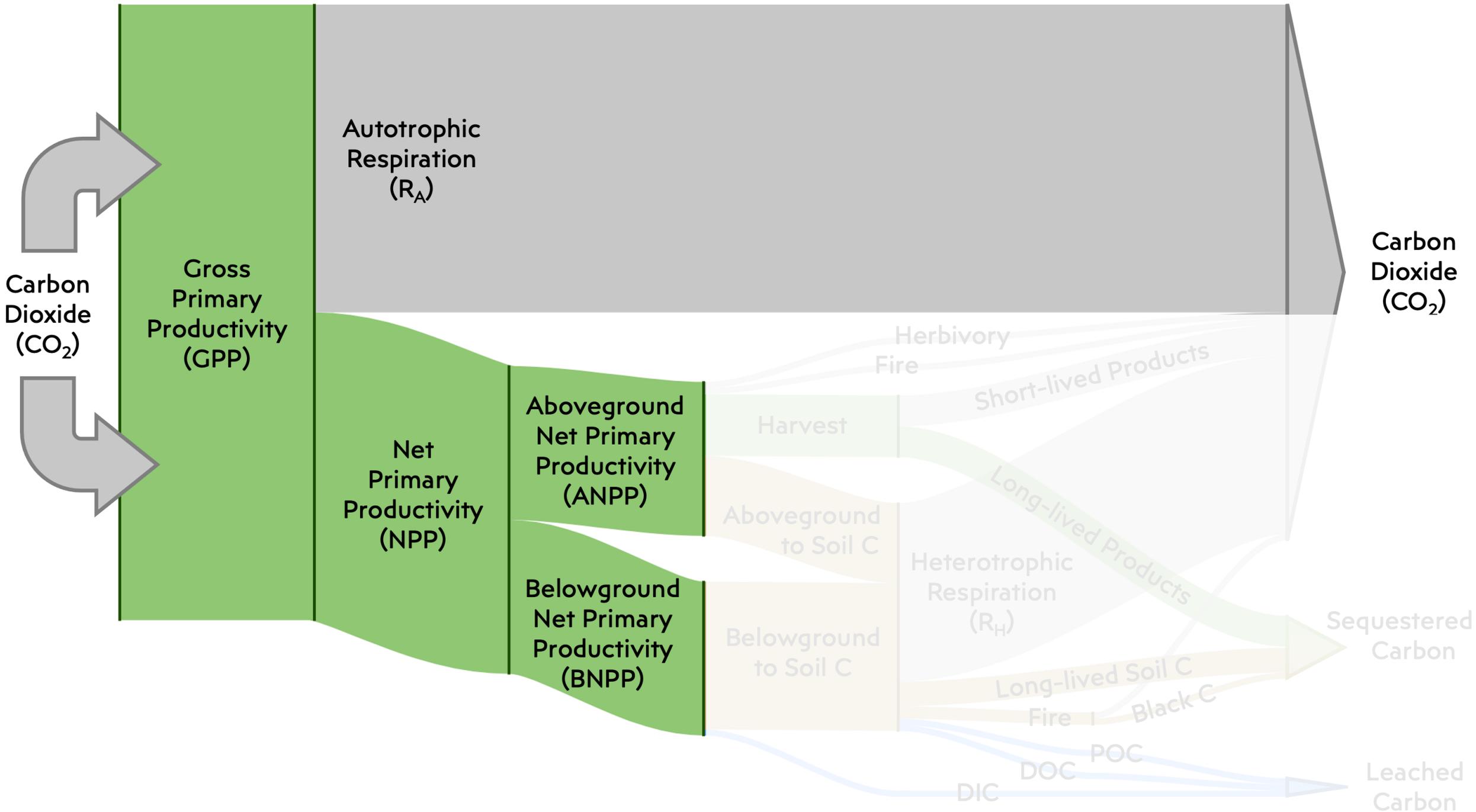


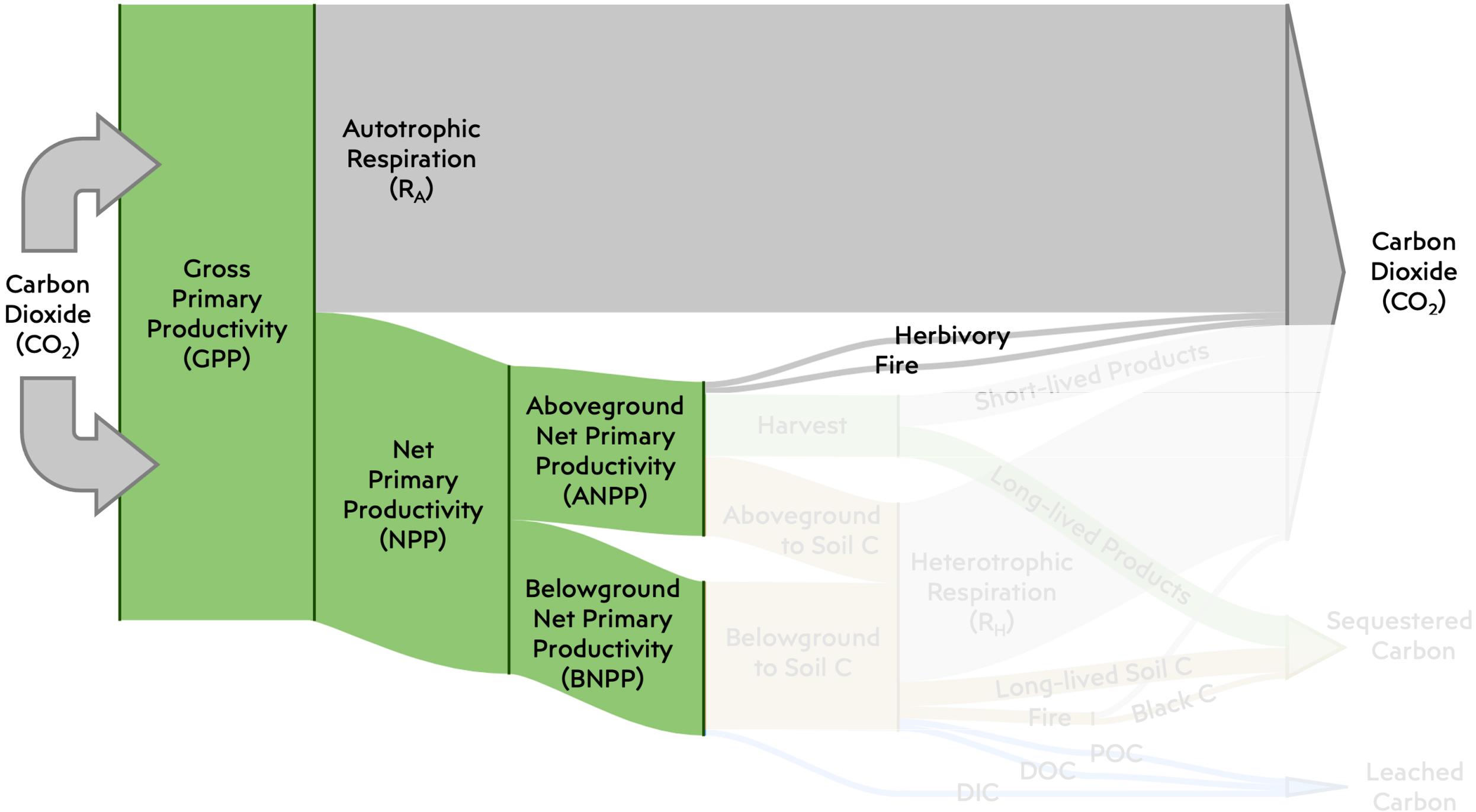


(adapted from Litton et al., 2007)

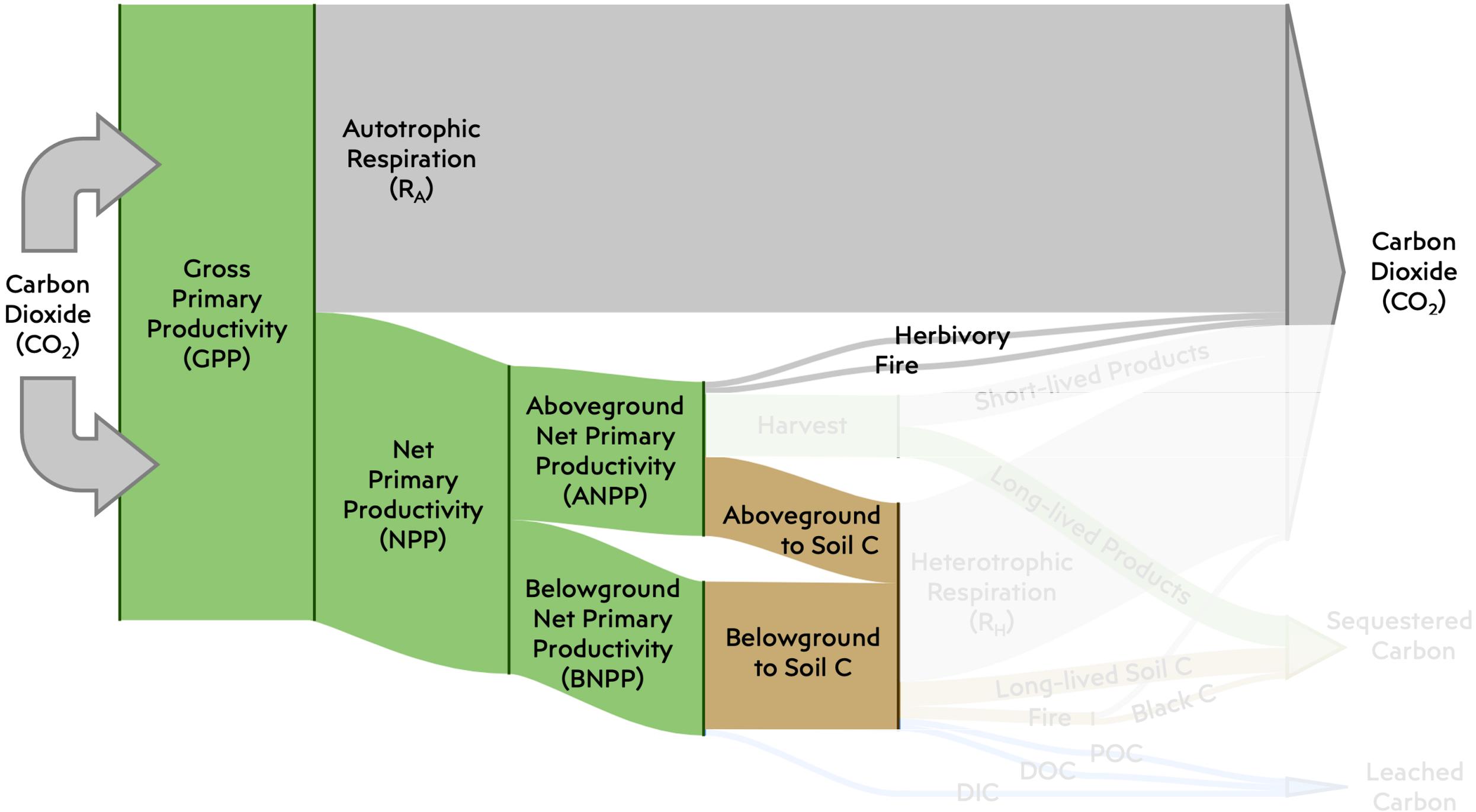


(Jiang and Wang, 2017)

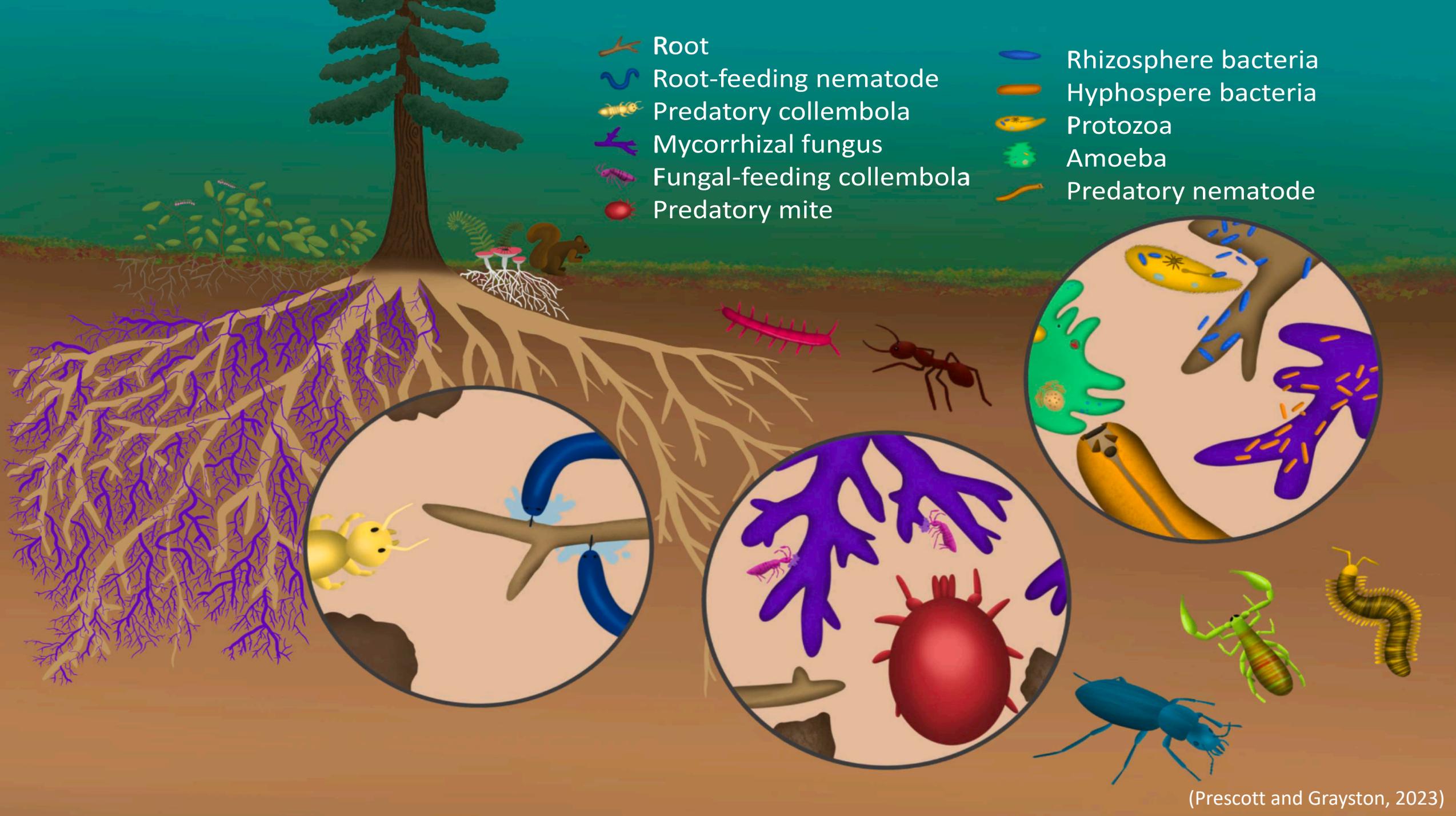






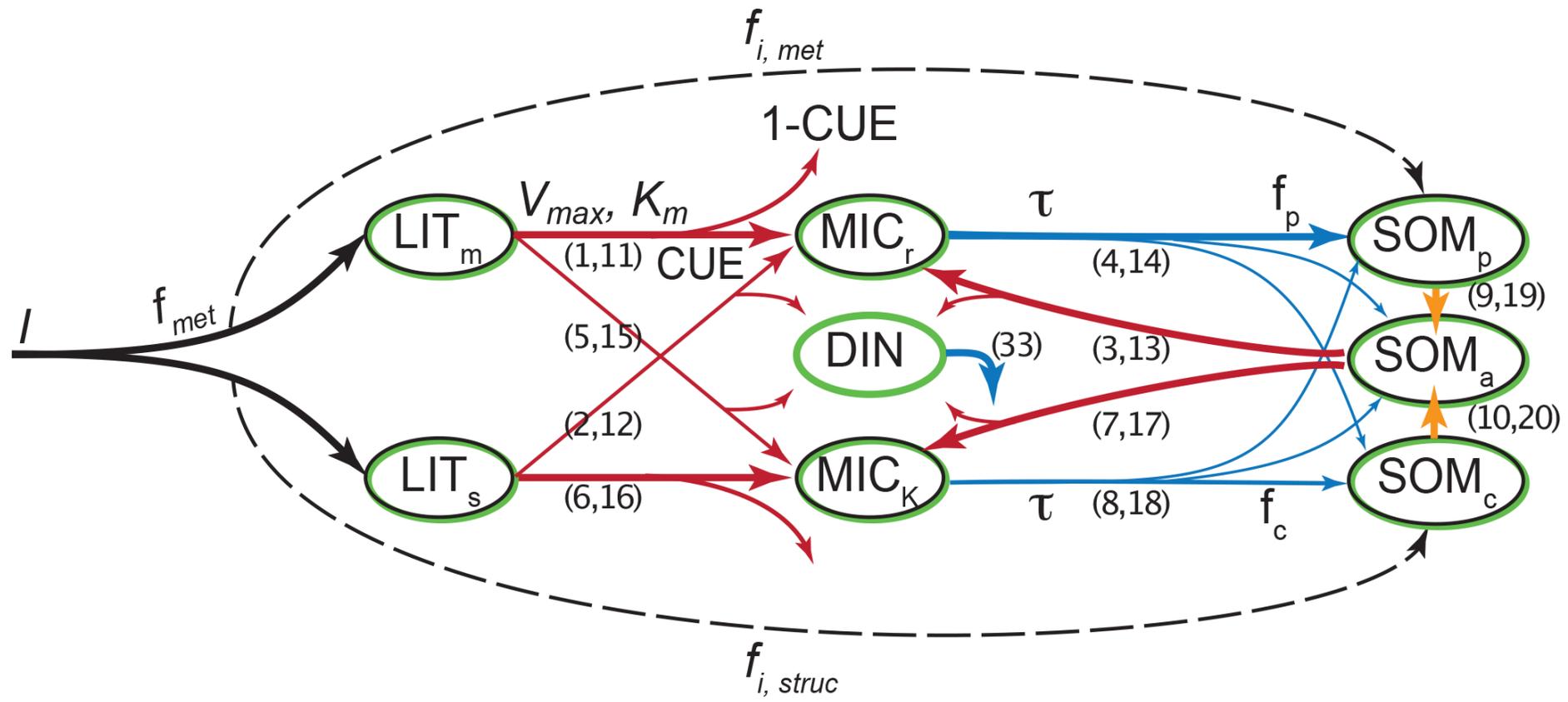


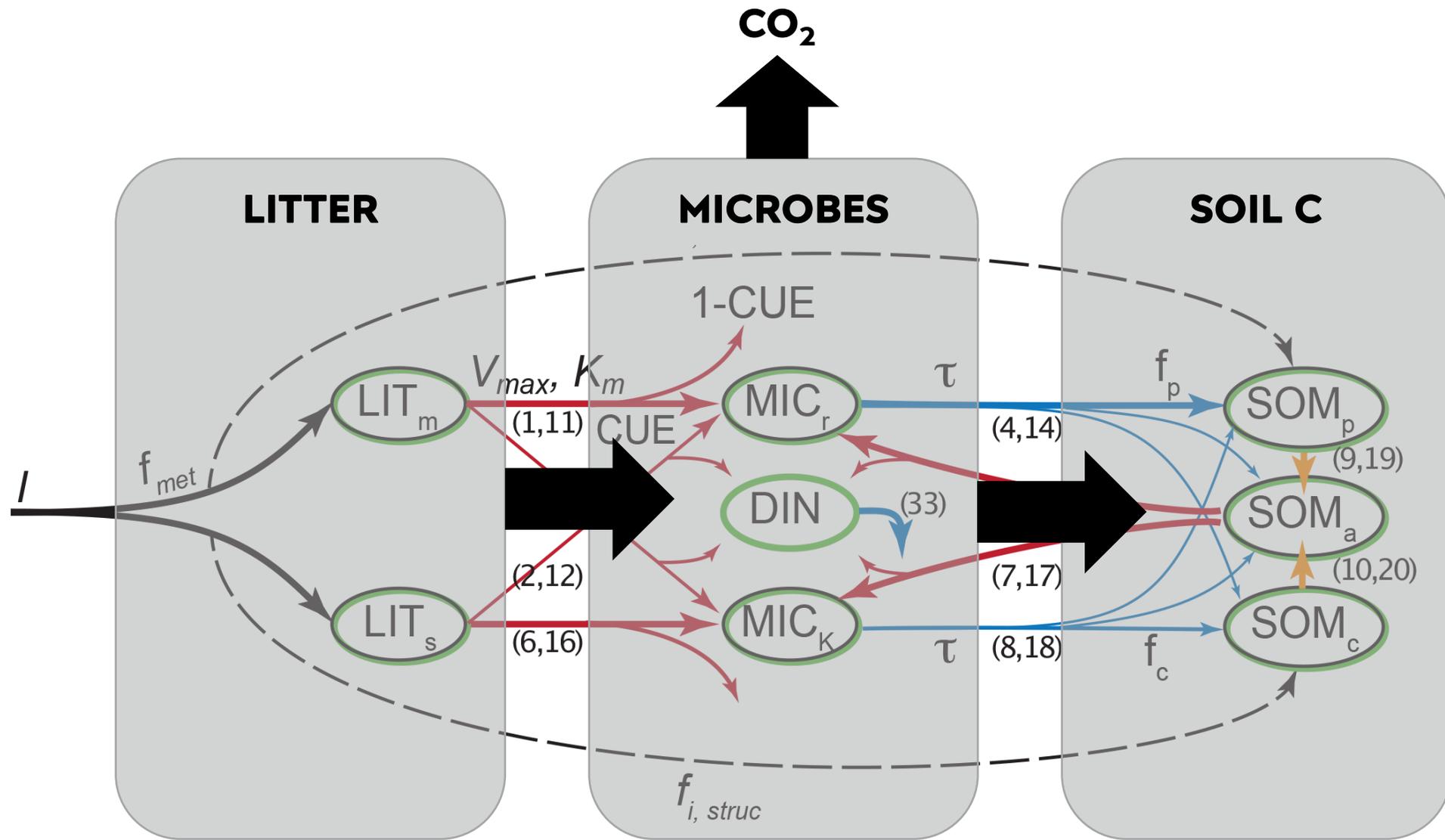


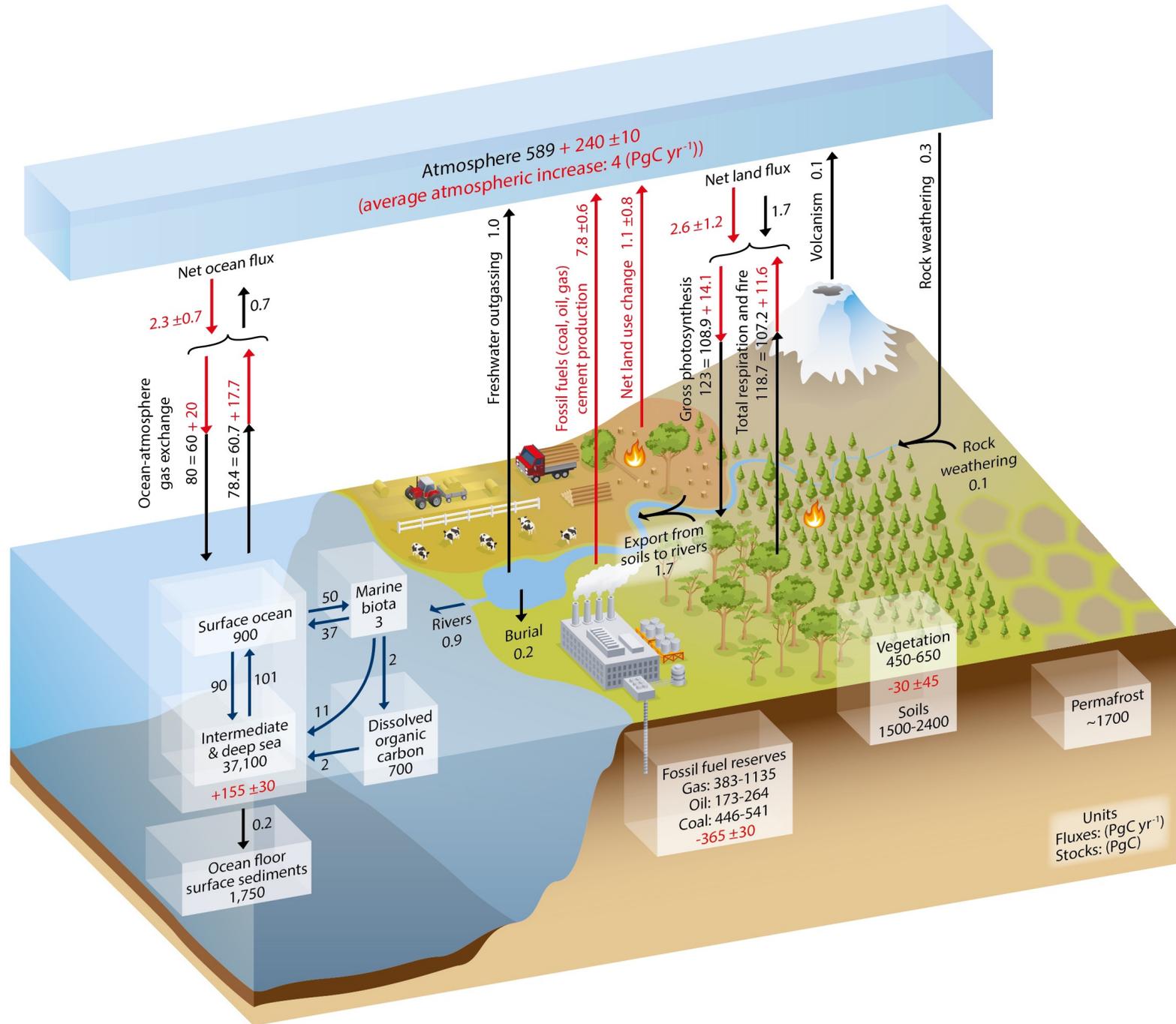


- Root
- Root-feeding nematode
- Predatory collembola
- Mycorrhizal fungus
- Fungal-feeding collembola
- Predatory mite

- Rhizosphere bacteria
- Hyphospere bacteria
- Protozoa
- Amoeba
- Predatory nematode









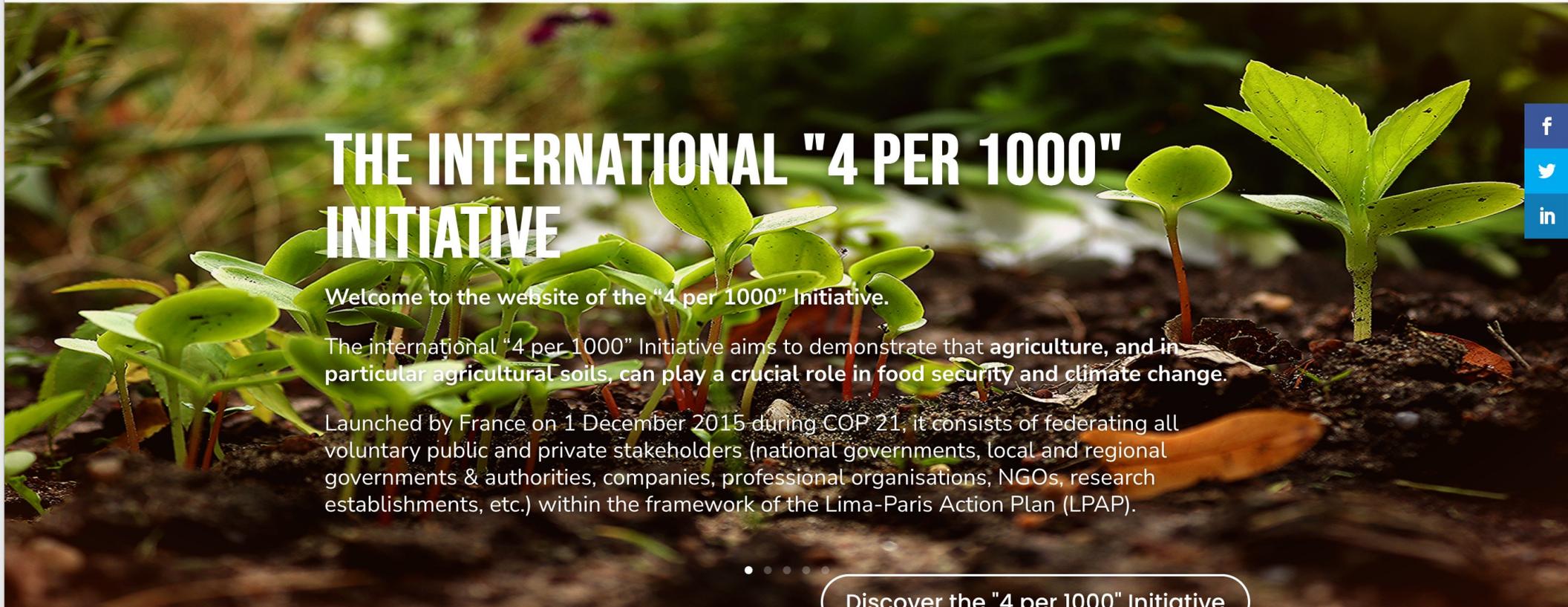
The international "4 per 1000" Initiative  
Soils for Food Security and Climate



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# THE INTERNATIONAL "4 PER 1000" INITIATIVE

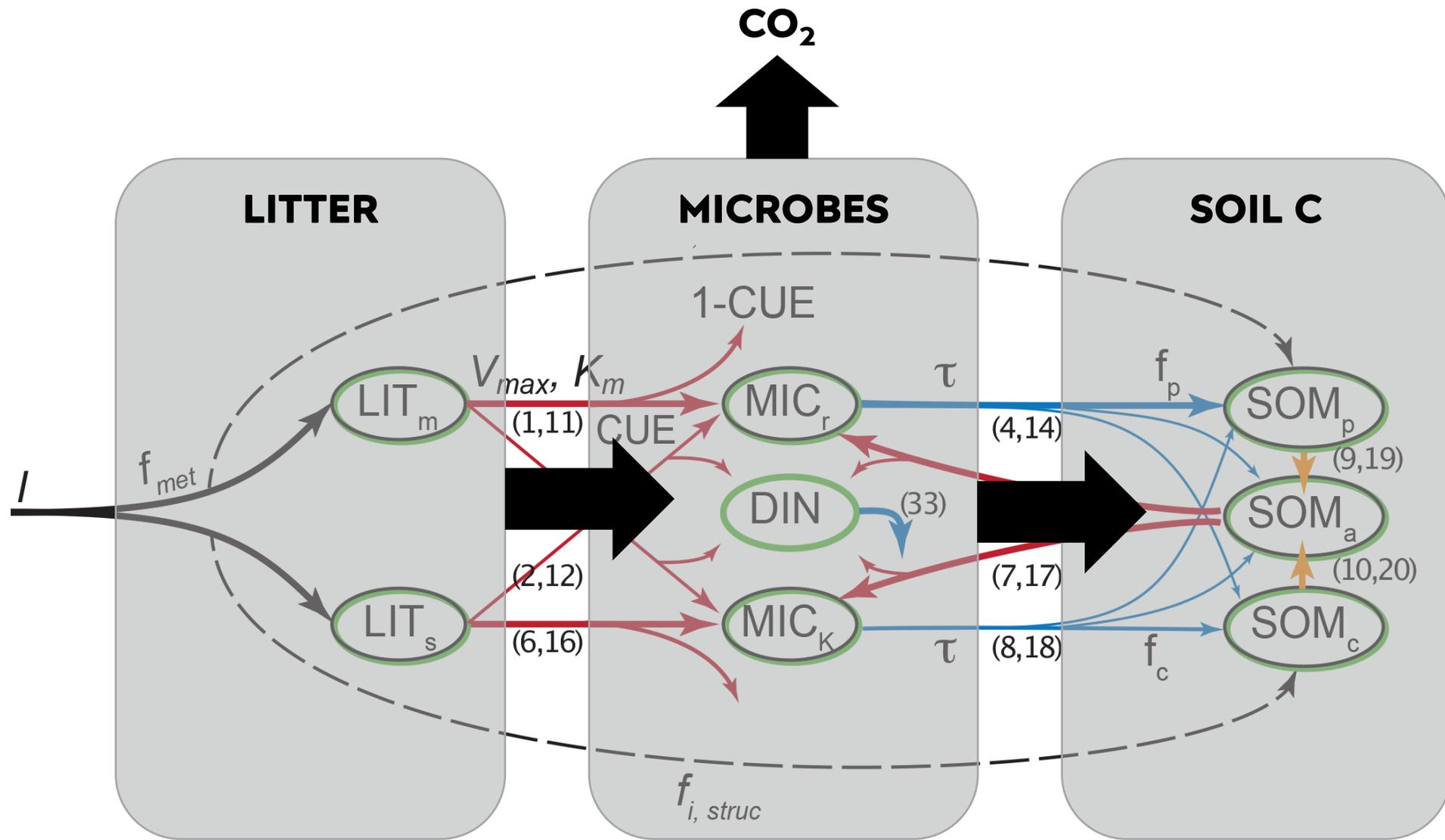
Welcome to the website of the "4 per 1000" Initiative.

The international "4 per 1000" Initiative aims to demonstrate that **agriculture, and in particular agricultural soils, can play a crucial role in food security and climate change.**

Launched by France on 1 December 2015 during COP 21, it consists of federating all voluntary public and private stakeholders (national governments, local and regional governments & authorities, companies, professional organisations, NGOs, research establishments, etc.) within the framework of the Lima-Paris Action Plan (LPAP).

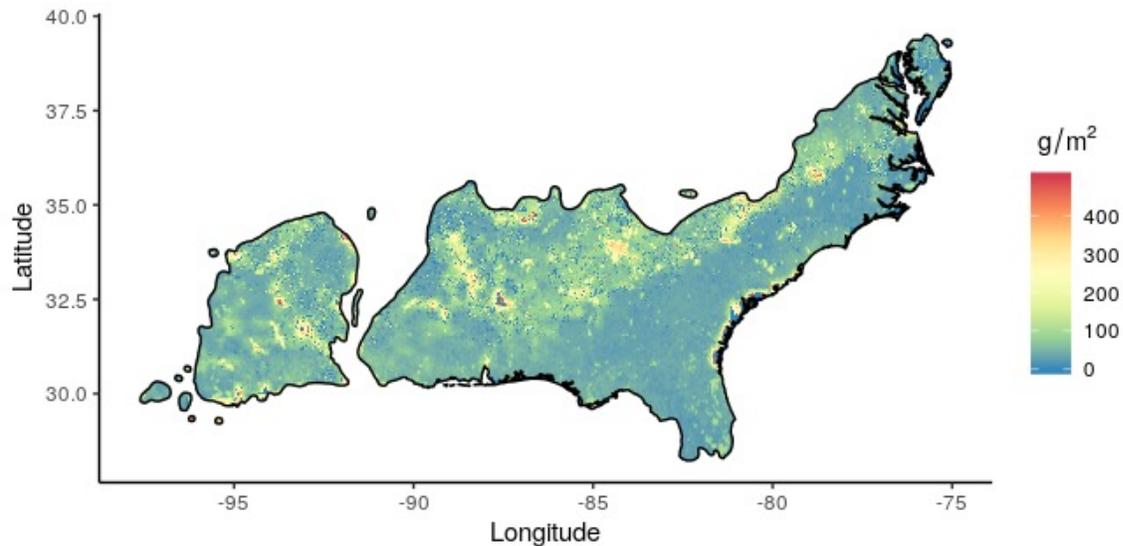
[Discover the "4 per 1000" Initiative](#)

## UNDERSTANDING THE "4 PER 1000" INITIATIVE IN 3'30

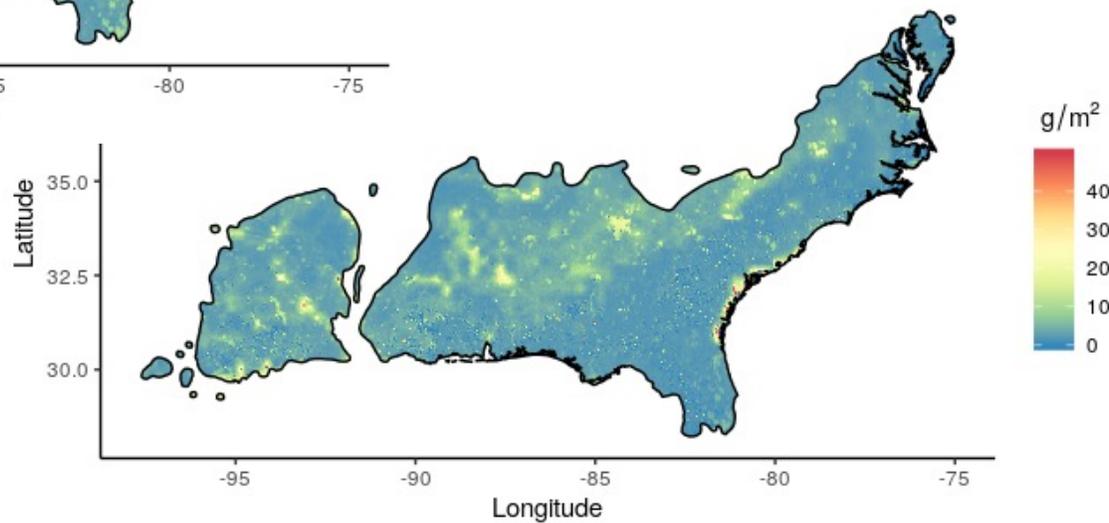




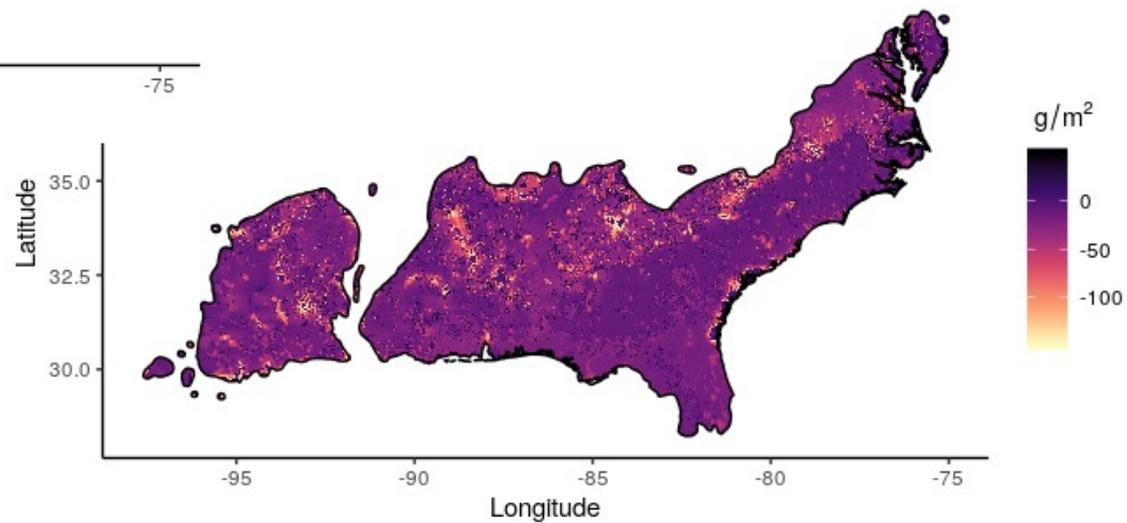
# Current climate

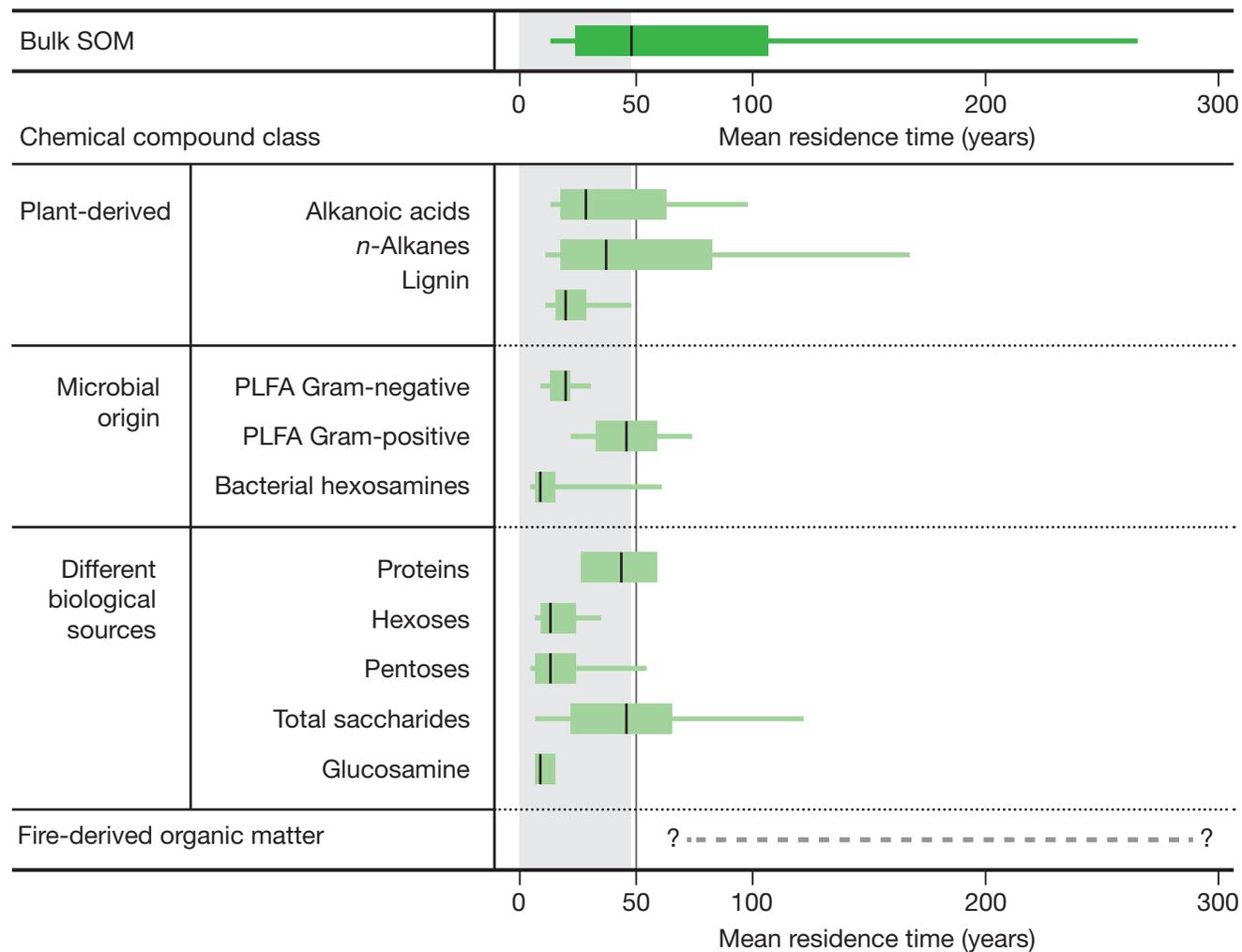


# Future climate

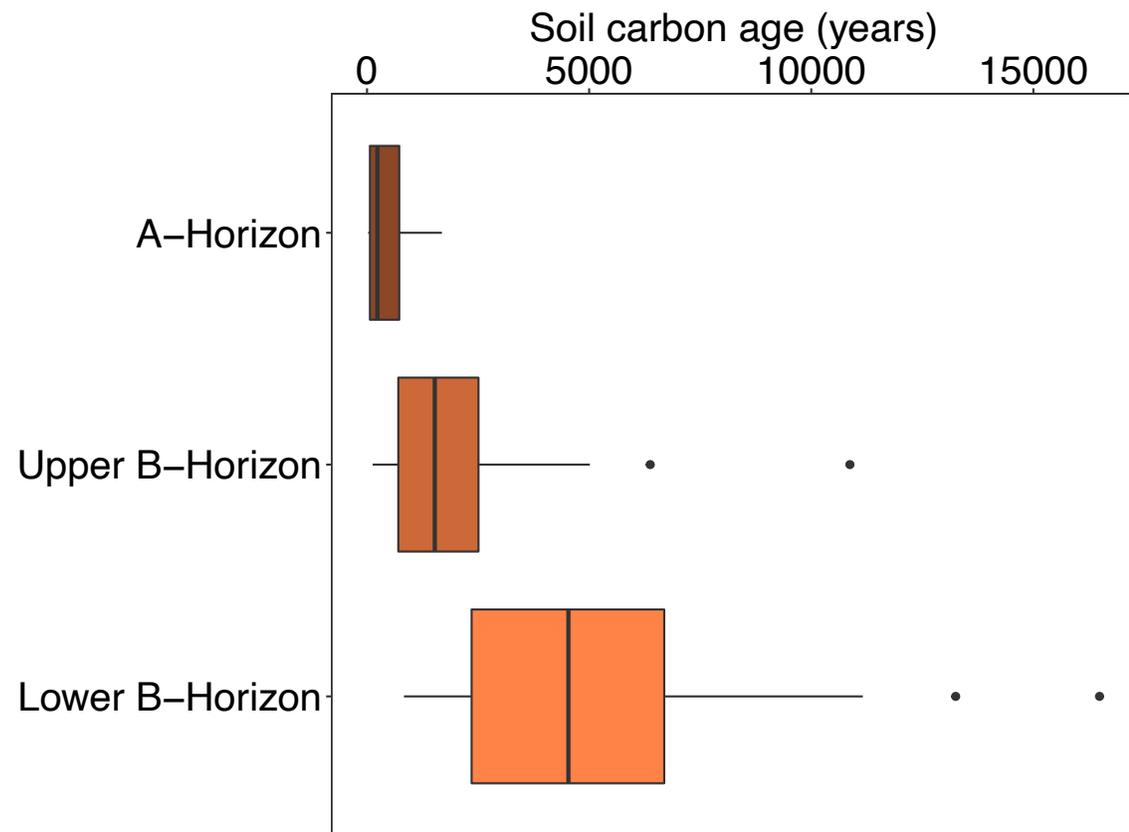


# Soil C change

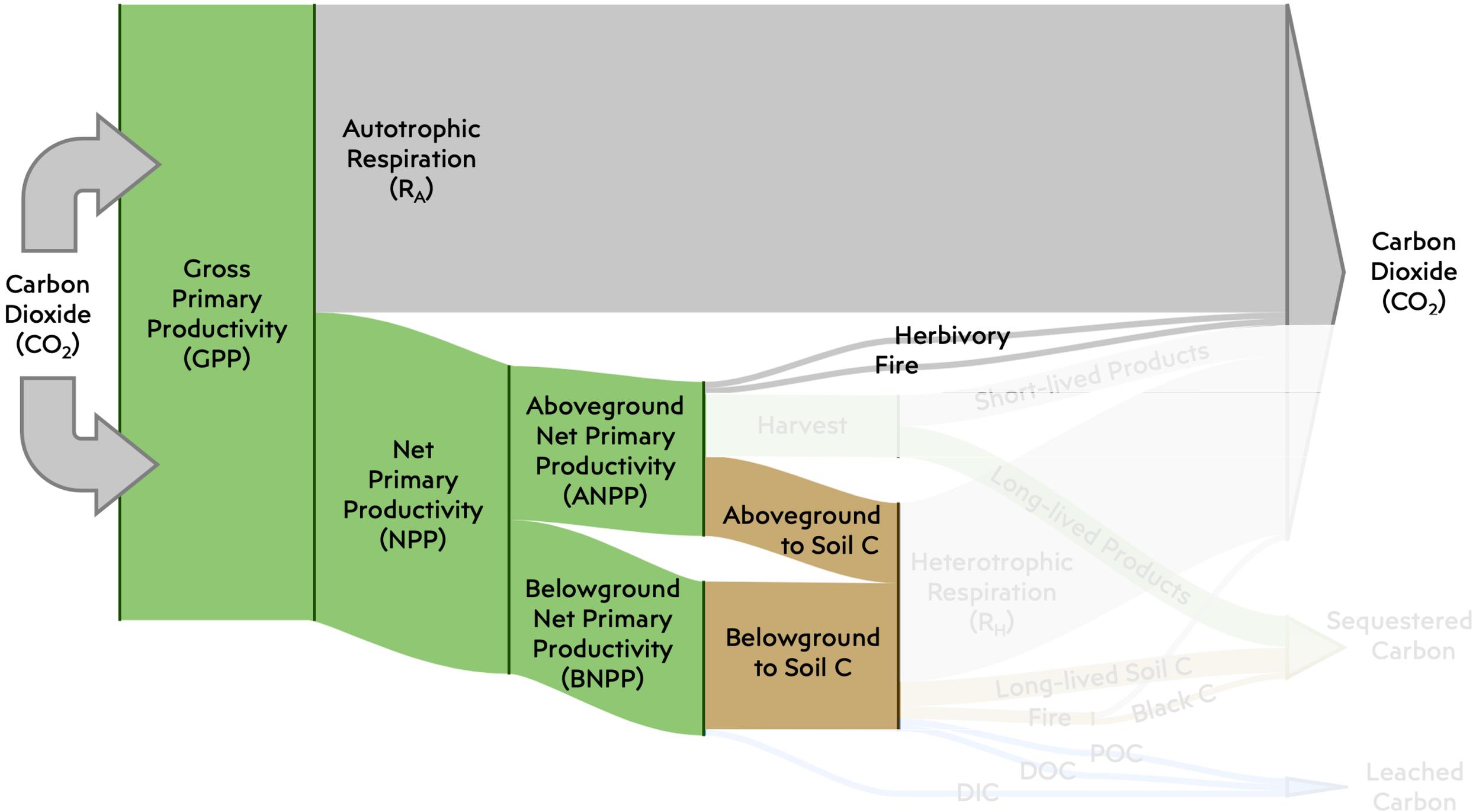


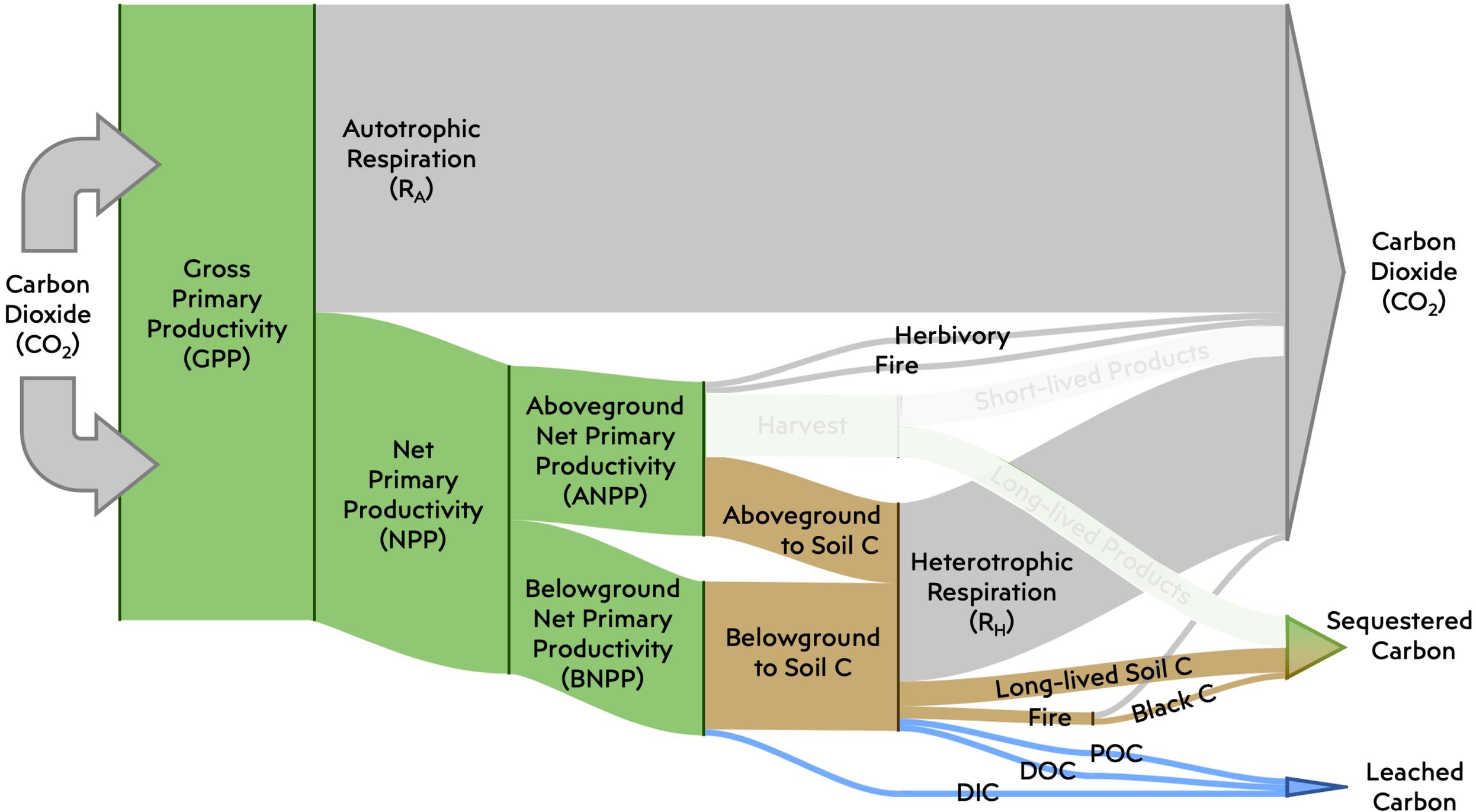


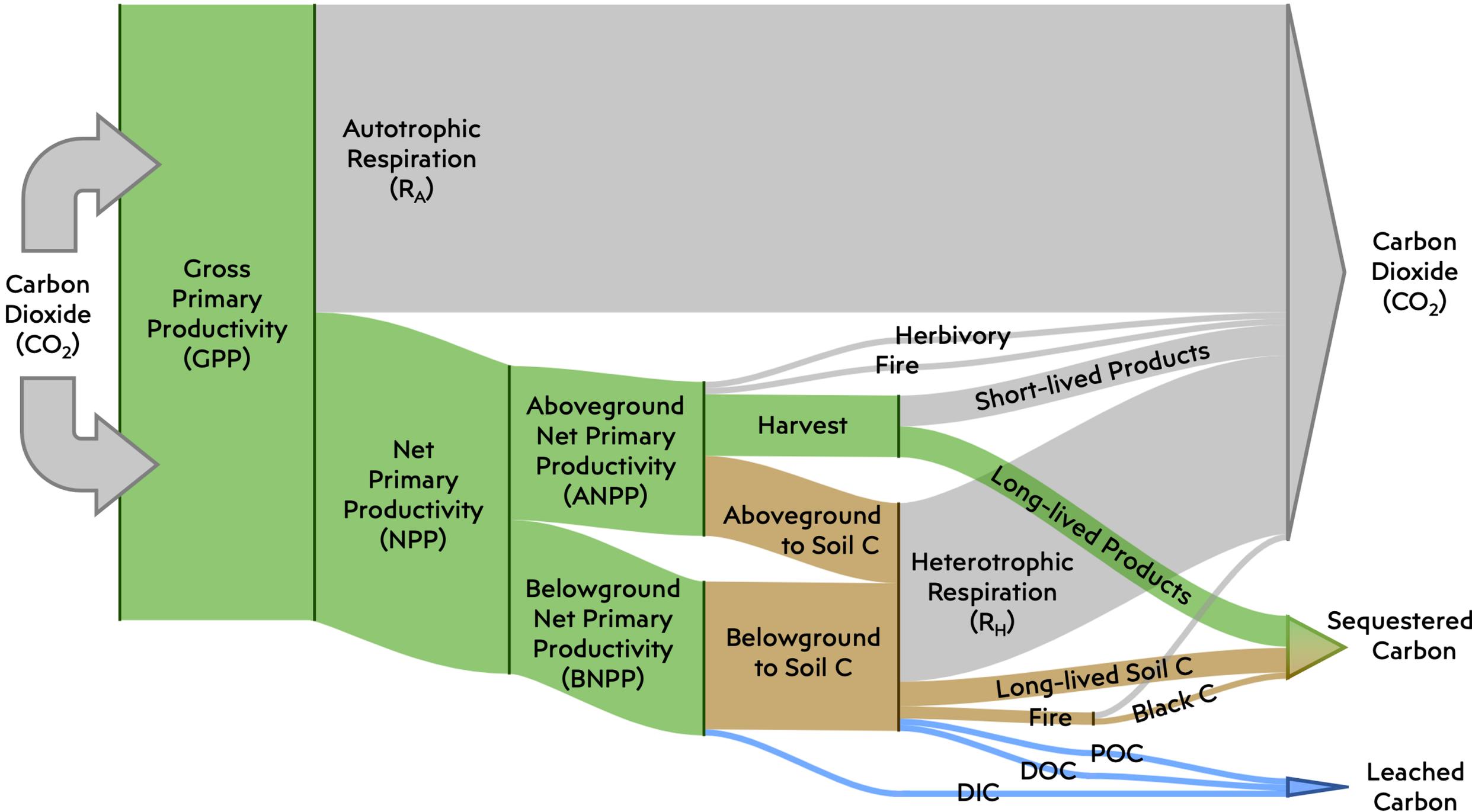
(Schmidt et al., 2011)



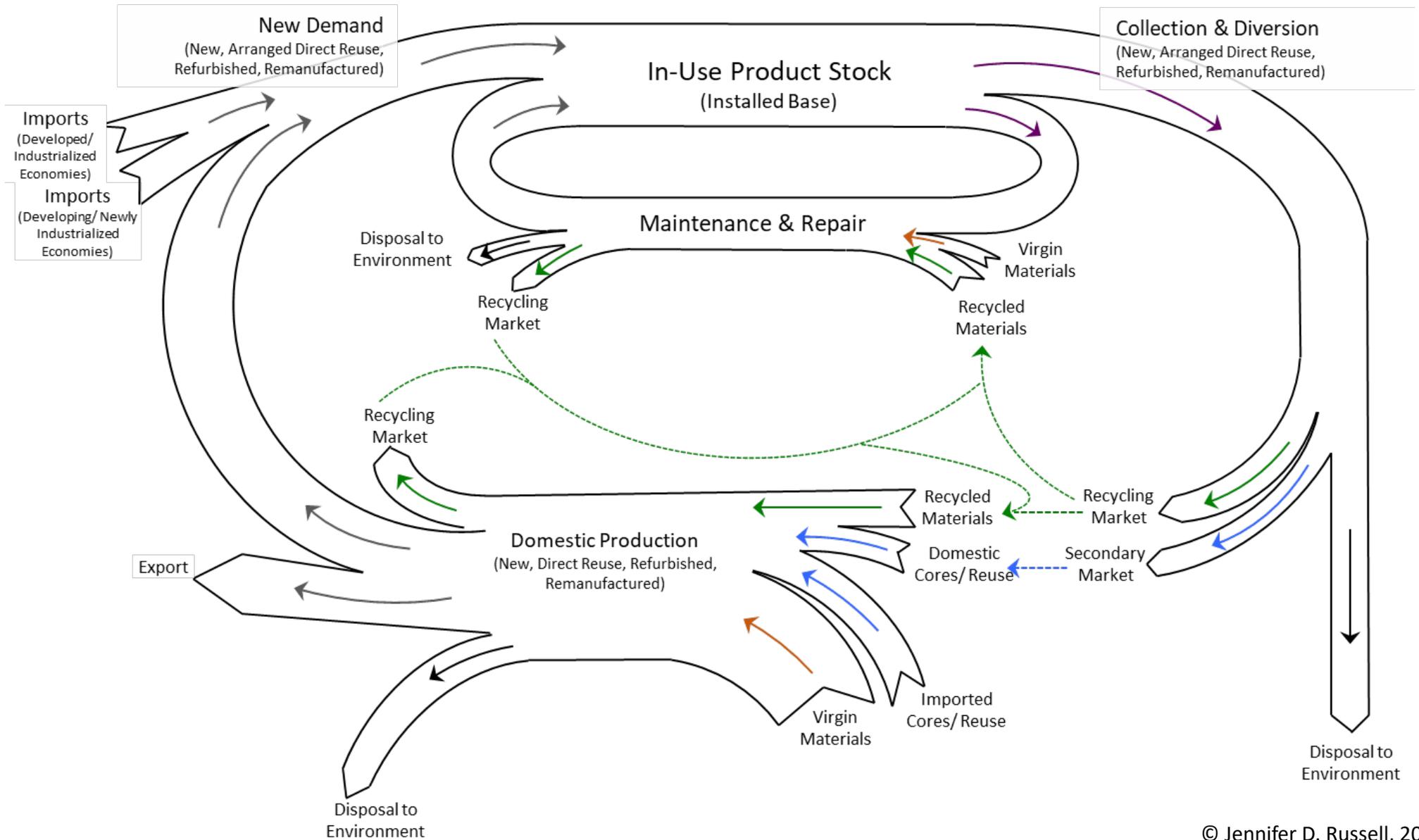
(National Ecological Observatory Data  
from Strahm et al., 2023)







**Dr. Jennifer Russell**  
 Dept. of Sustainable Biomaterials  
 College of Natural Resources and Environment  
 Virginia Tech



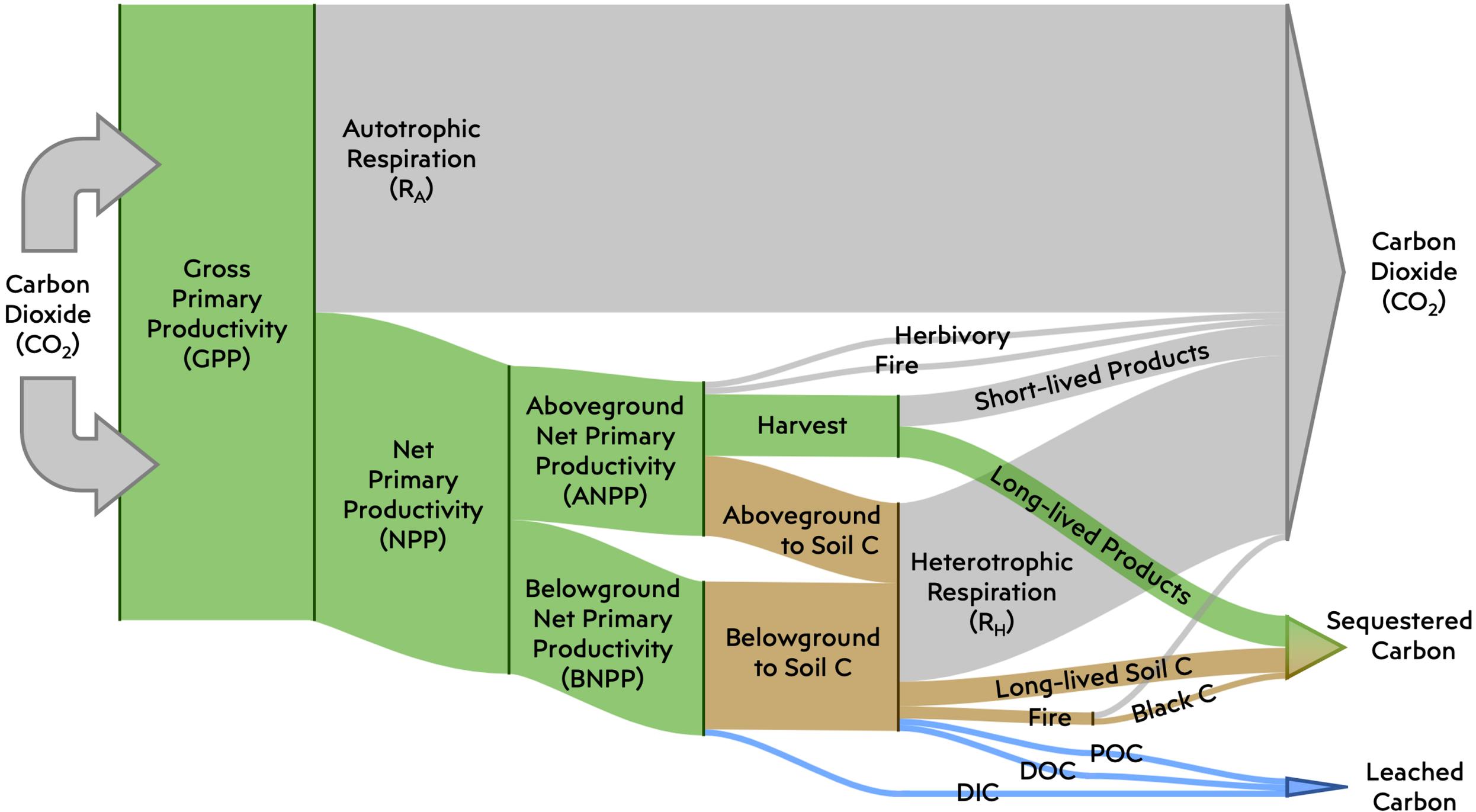
## Forest management and carbon sequestration in wood products

Ingolf Profft · Martina Mund · Georg-Ernst Weber ·  
Eberhard Weller · Ernst-Detlef Schulze

**Table 7** Average mean residence time (MRT), and average total residence time of wood products from timber harvested in Thuringian state forests (this study) and of dead wood in temperate forest ecosystems (Wirth et al. 2004)

Species group	MRT (total residence time) (years)	
	Wood products	Dead wood
Beech	19 (56)	Deciduous trees, 14 (41)
Oak	20 (59)	
Spruce	21 (62)	Coniferous trees, 34 (103)
Pine	24 (73)	
Mean Thuringia <sup>a</sup>	20 (62)	28 (84)

<sup>a</sup> Weighted by area. MRT =  $t_{63}$  = 63% of the initial amount of a wood product is decomposed or burned, total residence time =  $t_{95}$  = 95% of the initial amount of a wood product is decomposed or burned







# Carbon: Seeing the Forest (and Products) for the Trees

**Brian D. Strahm**

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and Environmental Conservation

College of Natural Resources  
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Virginia Tech